# The Composition of Cigarette Smoke. An Historical Perspective of Several Polycyclic Aromatic Hydrocarbons\*

by

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#### SUMMARY

Because of the significant advancements in fractionation, analytical, and characterization technologies since the early 1960s, hundreds of components of complex mixtures have been accurately characterized without the necessity of actually isolating the individual component. This has been particularly true in the case of the complex mixtures tobacco and tobacco smoke. Herein, an historical account of a mid-1950 situation concerning polycyclic aromatic hydrocarbons (PAHs) in cigarette smoke is presented. While the number of PAHs identified in tobacco smoke has escalated from the initial PAH, azulene, identified in 1947 to almost 100 PAHs identified by late 1963 to more than 500 PAHs identified by the late 1970s, the number of PAHs isolated individually and characterized by several of the socalled classical chemical means (melting point, mixture melting point, derivative preparation and properties) in the mid-1950s and since is relatively few, 14 in all. They were among 44 PAHs identified in cigarette mainstream smoke and included the following PAHs ranging from bicyclic to pentacyclic: Acenaphthylene, 1,2-dihydroacenaphthylene, anthracene, benz[a]anthracene, benzo[a]pyrene, chrysene, dibenz[a,h]anthracene, fluoranthene, 9H-fluorene, naphthalene, 1-methylnaphthalene, 2-methylnaphthalene, phenanthrene, and pyrene. One of them, benzo[a]pyrene,was similarly characterized in another study in 1959 by HOFFMANN. [Beitr. Tabakforsch. Int. 23 (2009) 384–410]

Wegen bedeutender Fortschritte bei der Fraktionierung, der Analytik und der Charakterisierung von Substanzen seit den frühen 1960iger Jahren wurden Hunderte von Verbindungen komplexer Gemische präzise charakterisiert, ohne dass die Notwendigkeit bestand einzelne Substanzen zu isolieren. Dies gilt in besonderem Maße für die komplexen Gemische Tabak und Tabakrauch. Diese Arbeit beschreibt in einem historischen Rückblick den Wissenstand zu den polyzyklischen aromatischen Kohlenwasserstoffe (PAHs) im Zigarettenrauch in der Mitte der 1950iger Jahre. Während die Anzahl der im Tabakrauch identifizierten PAHs vom zuerst bekannten Azulene, das 1947 identifiziert wurde auf fast 100 PAHs am Ende des Jahres 1963 und mehr als 500 PAHs am Ende der 1970iger Jahre stark anwuchs, ist die Anzahl der individuell analysierten und durch die Methoden der sogenannten klassischen Chemie charakterisierten (Schmelzpunkt, Siedepunkt, Derivate und deren Eigenschaften) seit der Mitte der 1950iger Jahre bis heute relative gering, insgesamt 14. Diese PAHs befinden sich unter den 44 PAHs, die sich im Hauptstromrauch von Zigaretten befinden und umfassten die nachfolgenden bizyklischen bis pentazyklischen PAHs: Acenaphthylen, 1,2-Dihydroacenaphthylen, Anthracen, Benz[a]anthracen, Benzo[a]pyren, Chrysen, Dibenz[a,h]anthracen, Fluoranthen, 9H-Fluoren, Naphthalin, 1-Methylnaphthalin, 2-Methylnaphthalin, Phenanthren und Pyren. Eine dieser Substanzen, und zwar Benzo[*a*]pyren, wurde außerdem in einer anderen Arbeit im Jahre 1959 von HOFFMANN charakterisiert. [Beitr. Tabakforsch. Int. 23 (2009) 384-410]

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<sup>\*</sup>Received: 9th September 2008 - accepted: 4th May 2009

#### RESUME

A cause des progrès significatif dans les technologies de fractionnement, d'analyse et de caractérisation depuis le début des années 1960, des centaines de composants de mélanges complexes ont été caractérisés de façon précise sans pour autant devoir isoler le composent individuel. Ceci a particulièrement été le cas pour les mélanges complexes du tabac et de la fumée de tabac. Cet étude présente la situation des hydrocarbures polynucléaires aromatiques (PAH) dans la fumée de cigarettes au milieu des années 1950. Tandis que le nombre des PAHs identifiés dans la fumée de tabac a rapidement augmenté, du PAH initial, l'azulène, identifié en 1947, à presque 100 PAHs identifiés à la fin de l'année 1963, pour arriver à plus de 500 PAHs identifiés à la fin de l'année 1970, le nombre des PAHs isolés individuellement et caractérisés par plusieurs des moyens chimiques appelés classiques (point de fusion, point de fusion du mélange, propriétés et préparation de dérivés) au milieu des années 1950 et depuis cette epoche, est resté relativement bas, 14 en tout. Ceux-ci sont parmi les 44 PAHs identifiés dans la fumée principale de cigarette et comprennent les PAHs suivants, allant des PAHs bicycliques aux PAHs pentacycliques. acenaphthylène, 1,2dihydroacénaphthylène, anthracène, benz[a]anthracène, benzo[a]pyrène, chrysène, dibenz[a,h]anthracène, fluoranthène, 9H-fluorène, naphthalène, 1-methylnaphthalène, 2-methylnaphthalène, phenanthrène, and pyrène. Un des PAHs a parallèlement été caractérisé en 1959 dans une autre étude par HOFFMANN. [Beitr. Tabakforsch. Int. 23 (2009) 384–410]

#### INTRODUCTION

Over fifty years ago, by use of the so-called classical chemistry method, 14 polycyclic aromatic hydrocarbons (PAHs) were isolated from the cigarette smoke condensate (CSC) of a commercial cigarette. Twelve of the 14 PAHs were isolated in crystalline form and two as oils. Although complete R.J. Reynolds Tobacco Company (RJRT) in-house reports describing the research (1, 2, 3) are available on the Internet <sup>a</sup> and a summary of the research was included in a recent publication (4), the complete details of it have never been published.

Between the early 1950s and the mid-1960s, numerous reports were published on the identification of PAHS in tobacco smoke, particularly cigarette smoke. In most of the pre-1957 studies, the evidence presented for the identification of a particular PAH consisted solely of ultraviolet (UV) absorption spectral data. The early claims to the identification of benzo[a]pyrene (B[a]P) in cigarette smoke (5, 6, 7), based on UV spectral data, were criticized by other investigators because of a) their inability to duplicate the reported experiments by use of similar techniques (8) or b) because of the incompleteness of the UV spectral data (9). However,

subsequent study (10) or assessment of more recent data (11) by the critics subsequently resulted in reversal of their original opinion.

Since 1957, numerous technologies have been used to isolate and identify trace amounts of certain PAHs in tobacco smoke. These include: Fluorescence spectrometry, paper chromatography, thin-layer and vapor-phase chromatography, isotope dilution, liquid-liquid partition, and mass spectrometry. The advantages and disadvantages of many of these technologies were discussed at length by WYNDER and HOFFMANN in their 1964 review (12) and 1967 book (13) on tobacco and tobacco smoke. In 1959, WYNDER and HOFFMANN reported the isolation of B[a]P in crystalline form from CSC and the establishment of its identity by evidence other than UV spectral data, namely, by melting point comparison and mixture melting point data with an authentic B[a]P sample (14). Because the study described herein was conducted in the mid-1950s but never published as such, the similar criteria used to identify PAHs isolated in crystalline or oil form from CSC might be of some historical interest to the reader.

To maintain chronological perspective, most references utilized in the discussion on PAHs herein will precede the issuance of the 1964 smoking-health report of the Advisory Committee to the U.S. Surgeon General (15). At that time, 97 PAHs had been identified in CSC, 18 of which were noted specifically by the Advisory Committee, 61 by Philip Morris in its 1963 monograph on tobacco and tobacco smoke components (16) (provided in 1963 to the Advisory Committee), and 77 by R.J. Reynolds (1, 2, 17) in various research reports [see Table 2 in RODGMAN (4)]. In a manuscript submitted for publication in late 1963 and published in early 1964 [shortly after release of the Advisory Committee report (15)], ELMENHORST and RECKZEH (18) catalogued 89 aromatic hydrocarbons identified in tobacco smoke. Of the 89, 77 were PAHs, the other 12 were monocyclic aromatic hydrocarbons identified in tobacco smoke. Obviously, the Advisory Committee did not have access to this manuscript prior to the publication of its report to the U.S. Surgeon General. However, it is an indication that the author of the tobacco and tobacco smoke chapter in the 1964 Advisory Committee was much less meticulous in collecting published tobacco smoke PAH data than were ELMENHORST and RECKZEH.

With the advent of many new fractionation and analytical technologies and particularly their excellent utilization by SNOOK *et al.* (19) at the U.S. Department of Agriculture (USDA), the number of PAHs identified in CSC underwent a more than five-fold increase between late 1963 and the late 1970s [see Table 6 in RODGMAN and PERFETTI (20)].

The fractionation-isolation procedure used in this mid-1950 CSC PAH study at RJRT Research Department was one originally designed in the early 1950s by one of us (A.R.) for a research study conducted at the Banting and Best Department of Medical Research at the University of Toronto, Toronto, Ontario, Canada on an industrial tar, one generated during an industrial process in a major U.S. industry. Two existing technologies were combined: a) The ability of PAHs to form complexes with picric acid, the technology used in the early 1930s by COOK *et al.* (21) in their isolation and identification of several PAHs, including B[a]P and benzo[e]pyrene (B[e]P), from coal tar. b) The

<sup>&</sup>lt;sup>a</sup> While the date on the in-house RODGMAN-COOK report (3) is mid-1960, the isolation/identification research on the 44 PAHs was actually completed between mid-1956 and early 1958. The delay in finalizing the report was due to personnel involvement in a detailed study of the effect of organic solvent extraction of tobacco on the yield of various PAHs in cigarette mainstream smoke.

decomposition of the PAH:picric acid complexes during column chromatography on alumina and the successive elution of the PAHs by solvents such as pentane or hexane. In the first isolation study of PAHs from CSC, picric acid was used as the complexing agent. For a second, more extensive study, the well-known PAH complexing agent 2,4,7-trinitrofluorenone (TNF) (22) was substituted for the picric acid.

Two early studies on the identification of PAHs in mainstream CSC were conducted at RJRT, the first beginning in late 1954. The initial RJRT PAH investigation involved isolation and identification of 11 PAHs in the mainstream CSC from 10500 commercial non-filtered 70mm cigarettes (1). Of the 11 PAHs, five were isolated in crystalline form (see Table 1). The cigarettes were smoked in a system modeled after that described by WYNDER et al. in their generation of cigarette tar for their study of its carcinogenicity (23). Instead of the smoking procedure described in 1936 by BRADFORD et al. (24), i.e., 35-mL puff, 2-sec puff duration, one puff/min, the smoking of the cigarettes paralleled that described by WYNDER et al. (23), i.e., 35-mL puff, 2-sec puff duration, three puff/min. The butt length for non-filtered cigarettes was 20-25 mm; that for filtered cigarettes was the 17 mm of the filter tip plus 3-4 mm of the tobacco rod. This was done in order to generate a comparison between our PAH data with the biological data reported by WYNDER et al. (23). Certain fractions of this CSC were fractionated to isolate and characterize unequivocally by classical chemical methods some of the PAHs in the CSC. The isolation procedure involved the following sequence: Liquid-liquid partition, repeated column chromatography on alumina to obtain a PAH-rich fraction, treatment of the PAH-rich fraction with TNF to form PAH:TNF complexes, separation of the complexes from extraneous aliphatic hydrocarbons and other long-chained neutral components, decomposition of the PAH:TNF complexes on alumina to concentrate the PAHs, and repeated chromatography of the PAH mixture to yield individual PAHs.

Five PAHs – naphthalene, anthracene, pyrene, fluoranthene, and B[a]P – were isolated in crystalline form and characterized by UV spectral data as well as by the usual classical chemical means mentioned previously. The remaining six PAHs (phenanthrene, acenaphthylene, 2methylpyrene, 1,2-benz[a]anthracene (B[a]A), benzo[ghi]perylene, and dibenzo[def,mno]chrysene) were identified on the basis of agreement of their UV absorption spectra with those of authentic samples and/or with published UV spectral data.

The second investigation involved the mainstream CSC from 20000 commercial filter-tipped cigarettes (2). Identified were 44 PAHs <sup>b</sup>, including the 11 PAHs identified in the initial study (1). Of the 44 PAHs, 12 were isolated in crystalline form [acenaphthylene, 1,2-dihydro-acenaphthylene, anthracene, B[*a*]A, B[*a*]P, chrysene, dibenz[*a*,*h*]anthracene (DB[*a*,*h*]A), fluoranthene, 9*H*-

characterized not only by UV spectral data but also by other classical chemical means. Table 1 summarizes the characterization criteria used for these 14 PAHs plus estimates of their yields under the smoking conditions used in this study. The other 30 PAHs were identified primarily from the agreement of their UV absorption spectra with those of authentic samples or with published spectra (Table 2). B[a]P, B[a]A, and DB[a,h]A had been reported to be tumorigenic to mouse skin (25, 26) although the bioassay data on the tumorigenicity of B[a]A were contradictory. Several years after completion of the second study, two notable situations arose in which 1,2-dihydrobenz[j]aceanthrylene (cholanthrene) and dibenzo[def,p]chrysene (formerly named dibenzo[a,l]pyrene, initially 1,2,3,4dibenzopyrene or 3,4,8,9-dibenzopyrene) were involved. The changes over time in nomenclature for 660 PAHs are summarized in the 1997 National Institute of Standards and Technology publication authored by SANDERS and WISE (27). The original RODGMAN-COOK identifications of these two PAHs, based solely on UV spectral data, were later found to be incorrect. From UV spectral data that were not totally in agreement with those of an authentic sample, RODGMAN and COOK (3) incorrectly defined one of them as the PAH 1,2-dihydrobenz[*j*]aceanthrylene (cholanthrene). In the massive study by SNOOK et al. at the USDA on the identification of PAHs in cigarette smoke, 1,2-dihydrobenz[j]aceanthrylene was not among the several benzocyclopentanthracenes identified (19). As a result, the listing of 1,2-dihydrobenz[*j*]aceanthrylene (cholanthrene) has been changed from its original listing to benzocyclopentanthracene. The other incorrectly characterized PAH was dibenzo[def,p]chrysene. For its identification as a tobacco smoke component, not only RODGMAN and COOK (3) but also RODGMAN (2), BONNET and NEUKOMM (28), LYONS and JOHNSTON (29), LYONS (30), WYNDER and WRIGHT (31), and PYRIKI (32) relied on available UV spectral data, purportedly those of synthetic dibenzo[*def,p*]chrysene (dibenzo[*a*,*l*]pyrene). However, in 1966, LAVIT-LAMY and BUU-HOÏ (33) determined that the published UV spectral data were not those of dibenzo[a, l] pyrene but of its isomer dibenz[a, e] aceanthrylene (dibenzo[a,e]fluoranthene), generated during the supposed synthesis of dibenzo[*def,p*]chrysene (dibenzo[*a,l*]pyrene). As a result of the finding of LAVIT-LAMY and BUU-HOï, the entry of dibenzo[*def,p*]chrysene (dibenzo[*a,l*]pyrene) has been changed from its original listing to dibenz[a,e]aceanthrylene.

fluorene, naphthalene, phenanthrene, pyrene] and two as oils (1-methyl- and 2-methylnaphthalene) and all 14 were

## EXPERIMENTAL

All solvents employed in the chemical fractionation and column chromatography of the CSC fractions were optically transparent insofar as PAHs were concerned. The chromatographic adsorbents used were Aluminum Oxide Merck (Merck Cat. No. 71707) and Fluorisil® (60-100 mesh) from the Floridin Company, Tallahassee, FL. All melting and mixed melting points were determined with a Fisher-Johns melting point apparatus. UV absorption spectra were determined at 25 °C with a Beckman DK-2

<sup>&</sup>lt;sup>b</sup> In a summary in a previous publication (4), the number of PAHs reported was 43 instead of 44: 14 characterized by various aspects of classical chemistry, 29 characterized by UV absorption spectra. Inadvertently omitted from that manuscript was the identification of the PAH benz[e] acephenanthrylene (ben-zo[b]fluoranthene).

			Ъ,	PAH		TNF C	TNF complex		Estim	Estimated quantity
CAS No.	Polycyclic aromatic hydrocarbon	UVª	$\mathbb{R}^{a}$	d.m	m.m.p°	°.d.m	m.m.p. <sup>e</sup>	Other identification	Total isolated, mg	ug/kg of tobacco smoked
208-96-8	Acenaphthylene <sup>1</sup>	×	×	×	I	I	I	I	7.0	500
83-32-9	Acenaphthylene, 1,2-dihydro- <sup>f</sup>	×	I	×	×	×	۰ ۲	x picrate UV <sup>a</sup>	24.0	1720
120-12-7	120-12-7 Anthracene <sup>f.9</sup>	×	×	×	×	×	×	x picrate m.p. <sup>b</sup> x picrate IR <sup>a</sup>	3.85	275
								x picrate m.m.p. <sup>e</sup>		
56 55 3		>		>		>		x dione IR <sup>a</sup>	0.03	7
0-00-00		×	I	×	I	×	I		0.23	2
50-32-8	Benzo[ <i>a</i> ]pyrene <sup>f, g</sup>	×	×	×	×	I	I	x picrate m.p. <sup>b</sup>	0.90	64
								x picrate m.m.p. <sup>e</sup>		
218-01-9	Chrysene <sup>†</sup>	×	Ι	I	I	×	×	I	0.62	44
53-70-3	Dibenz[ <i>a</i> , <i>h</i> ]anthracene	×	I	I	I	I	I	x dione UV <sup>a</sup>	0.21	15
206-44-0	Fluoranthene <sup>f, g</sup>	×	×	×	×	×	×	I	3.45	260
86-73-7	9 <i>H</i> -Fluorene <sup>f</sup>	×	Ι	×	×	×	×	I	3.2	230
91-20-3	Naphthalene <sup>f, g</sup>	×	×	×	×	×	×	I	32.0	2280
90-12-0	Naphthalene, 1-methyl- <sup>f</sup>	×	Ι	I	I	×	I	I	5.0	360
91-57-6	Naphthalene, 2-methyl- <sup>f</sup>	×	Ι	I	I	×	I	I	14.0	1000
85-01-8	Phenanthrene <sup>f</sup>	×	×	×	×	×	×	I	1.95	140
129-00-0	Pyrene <sup>f, g</sup>	×	I	×	×	×	×	Ι	0.85	60

Table 1. Polycyclic aromatic hydrocarbons in cigarette smoke condensate identified by spectral plus other means

<sup>a</sup> Spectrum (UV, IR) of isolate was identical with that of an authentic sample and/or with that reported in the literature.

<sup>b</sup> Melting point (m.p.) of isolate was identical with that of an authentic sample and/or with that reported in the literature. <sup>c</sup> Mixture melting point (m.m.p.) of isolate with an authentic sample showed no depression.

<sup>d</sup> Melting point (m.p.) of TNF (2,4.7-trinitrofluorenone) complex of isolate was identical with that of an authentic sample and/or with that reported in the literature.

<sup>e</sup> Mixture melting point (m.m.p.) of TNF (or picric acid) complex of isolate with TNF (or picric acid) of an authentic sample showed no depression. <sup>f</sup>Spectral plus other identification data from RODGMAN and COOK (3), a study involving the fractionation of the MSS from 20000 filter-tipped cigarettes. <sup>g</sup>Spectral plus other identification data from RODGMAN (1), a study involving the fractionation of the MSS from 10500 non-filtered cigarettes. <sup>n</sup> UV and IR spectra of TNF complex of isolate were identical with those of an authentic sample of 1,2-dihydroacenaphthylene:TNF complex.

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		Estim	ated quantity
CAS No.	Polycyclic aromatic hydrocarbon	Total isolated, mg	µg/kg of tobacco smoked
779-02-2	Anthracene, 9-methyl-	0.60	43
205-99-2	Benz[e]acephenanthrylene	0.63	44
2319-96-2	Benz[a]anthracene, 5-methyl-	0.05	3
	Benzocyclopentanthracene <sup>a</sup>	0.09	6
7099-43-6	1 <i>H</i> -Benzo[ <i>a</i> ]cyclopent[ <i>h</i> ]anthracene, 2,3-dihydro-	~0.01	trace
7099-42-5	9 <i>H</i> -Benzo[ <i>a</i> ]cyclopent[ <i>i</i> ]anthracene, 10,11-dihydro-	~0.01	trace
203-12-3	Benzo[ <i>ghi</i> ]fluoranthene	0.03	trace
205-82-3	Benzo[/]fluoranthene	~0.01	trace
207-08-9	Benzo[k]fluoranthene	~0.01	trace
189-55-9	Benzo[ <i>rst</i> ]pentaphene	0.02	trace
191-24-2	Benzo[ghi]perylene	0.24	17
195-19-7	Benzo[c]phenanthrene		
25167-89-9	Benzo[a]pyrene, methyl-	0.08	6
192-97-2	Benzo[e]pyrene	0.62	44
41637-90-5	Chrysene, methyl-	0.30	22
191-07-1	Coronene	0.05	3
5385-75-1	Dibenz[ <i>a,e</i> ]aceanthrylene <sup>b</sup>	0.02	trace
189-64-0	Dibenzo[b,def]chrysene	~0.01	trace
191-26-4	Dibenzo[def,mno]chrysene	0.10	7
193-09-9	Dibenzo[de,qr]naphthacene	0.02	trace
30997-39-8	Fluoranthene, methyl-	0.60	43
4373-13-1	Naphthalene, 1,2-dihydro-4-methyl-	3.5	250
612-94-2	Naphthalene, 2-phenyl-	2.0	140
196-42-9	Naphtho[2,1,8-qra]naphthacene	0.02	trace
198-55-0	Perylene	0.20	14
29062-98-4	Phenanthrene, dimethyl-	2.0	140
883-20-5	Phenanthrene, 9-methyl-	0.85	60
2381-21-7	Pyrene, 1-methyl-	0.87	62
3442-78-2	Pyrene, 2-methyl-	0.05	3
3353-12-6	Pyrene, 4-methyl-	0.72	51

Table 2. Polycyclic aromatic hydrocarbons from cigarette smoke condensate characterized by ultraviolet absorption spectra	
only	

<sup>a</sup> Originally, because of the similarity but not total agreement of the UV spectrum of this PAH with that of 1,2-dihydrobenz[*j*]aceanthrylene (cholanthrene), RODGMAN and COOK (3) designated this PAH as 1,2-dihydrobenz[*j*]aceanthrylene (cholanthrene). The major reason for the change in designation was the following: Although numerous benzocyclopentanthracenes were identified in CSC by SNOOK *et al.* (19), none was designated as 1,2-dihydrobenz[*j*]aceanthrylene (cholanthrene).

<sup>b</sup> Dibenz[a,e]aceanthrylene, sometimes known as dibenz[a,e]fluoranthene, was originally classified as dibenzo[*def,p*]chrysene (dibenzo[*a,*]pyrene, 1,2,3,4-dibenzopyrene, 3.4.8.9-dibenzopyrene) in several CSC studies (2, 28, 29, 30, 31, 32) but the PAH originally designated as dibenzo[*a*,]pyrene from a synthesis was subsequently found by LAVIT-LAMY and BUU-HOI (33) to be its isomer, dibenz[*a*,*e*]aceanthrylene.

Ratio Recording Spectrometer, 10-mm silica cells, and PAH-free absolute ethanol or cyclohexane as solvent. Infrared (IR) absorption spectra were determined by John J. Whalen with a Perkin-Elmer Model 21 Spectrophotometer.

#### Smoking procedure

The 20000 cigarettes smoked and the smoking procedure were described in detail by RODGMAN and COOK (34). The weight of the tobacco smoked was 14.0 kg.

## Neutral-acidic fraction

The collection of the hexane-soluble neutral-acidic fraction (248 g) and its preliminary chromatography on alumina to yield a hexane-eluted fraction (38.6 g), a 4:1 hexane:benzene-eluted fraction (19.1 g), and a methanol-eluted fraction (113 g) were described previously (34).

## Chromatography of the hexane-eluted fraction

The hexane-eluted fraction (38.6 g) was chromatographed in 9- to 10-g lots on alumina [45 mm (diam)  $\times$  125 mm column] initially with hexane (100 mL) to remove the bulk of the saturated aliphatic hydrocarbons and neophytadiene (35). The next 175 mL of hexane yielded a non-fluorescent eluate (UV light exposure) which contained the squalenes and solanesenes (36). Hexane (8  $\times$  100 mL) and methanol (5  $\times$  100 mL) elution yielded highly fluorescent eluates of 0.6 to 0.8 g for each of the four chromatograms, totaling 3.2 g. This material when combined with the 4:1 hexane:benzene-eluted fraction (19.1 g) yielded 22.3 g of oil, highly fluorescent to UV light.

#### Complex formation with 2,4,7-trinitrofluorenone (TNF)

The fluorescent oil (22.2 g) in 25 mL of benzene at 60  $^{\circ}$  was treated with a solution of TNF (7.5 g) in glacial acetic acid (180 mL) [cf. (22)]. After heating the resulting solution to the boiling point, then allowing it to cool to 25  $^{\circ}$ C, a reddish-

brown precipitate formed. The suspension was concentrated to dryness, diluted with hexane (200 mL), agitated for 0.5 h, and filtered to remove hexane-soluble long-chained compounds which did not complex with TNF. The residue, consisting of TNF:PAH complexes plus unreacted TNF, was washed with hexane (150 mL), filtered, and dried. Concentration of the filtrate and washings yielded an oil (20.4 g). Chromatography of a 1.0-g aliquant yielded no fraction with the characteristic blue fluorescence of the PAHs. Treatment of a similar aliquant with TNF (0.35 g) gave no indication of TNF:PAH complex formation.

#### Chromatography of the TNF:PAH complex mixture

The TNF:PAH complex mixture was decomposed by chromatography on alumina [75 mm (diam)  $\times$  100 mm column] (22). The mixture was added to the column in 1:1 benzene:chloroform (150 mL) and the PAHs eluted with 8 L of 4:1 hexane:benzene. Concentration yielded 0.748 g of a PAH mixture which exhibited an intense-blue-green fluorescence in UV light. Per kg of tobacco smoke, the PAH mixture represented about 53 mg of PAHs.

The PAH mixture in 1:1 benzene:hexane (30 mL) on alumina [25 mm (diam)  $\times$  600 cm column] <sup>c</sup> was eluted by 325 100-mL solvent treatments. After development of the column, exposure to UV light indicated 17 distinct and separate fluorescent zones, ranging from pale blue to blueviolet in color. In many instances, diffuse fluorescent zones were present between the more distinctly fluorescent zones. Table 3 lists the eluting solvents used plus the approximate constitution of the fractions obtained. The constitution of some PAHs could not be determined from their UV absorption. As a result, they were not included in Table 3. Each fraction collected was concentrated, dissolved in absolute ethanol or cyclohexane, and its UV absorption spectrum determined. Based on UV absorption similarities, certain fractions were combined (see Table 3) and rechromatographed on alumina to separate the suspected PAHs. Some over-laps were inevitable between the various combined subgroups but they are not shown in Table 3.

To preserve remaining PAH fractions while a particular fraction was being investigated, the combined fractions containing particular groups of PAHs were treated with excess TNF in benzene or ethanol, concentrated to dryness, and the resulting TNF:PAH complex mixture stored in the dark. Subsequent decomposition of the TNF:PAH complexes on alumina permitted recovery of the PAH(s) for further study.

## Polycyclic aromatic hydrocarbons

The quantities of the various PAHs found in this study are minimal values. The identity of each of the various PAHs was based on one or more of the following: UV absorption<sup>d</sup>, IR absorption, melting point (m.p.) and/or mixture melting

points (m.m.p.) of the isolate and of a derivative [PAH:picric acid complex, TNF:PAH complex, quinone (dione)].

1) 1-Methyl-3,4-dihydronaphthalene (CAS No. 4373-13-1): This PAH was tentatively identified on the basis of its UV absorption spectrum. It was essentially the same as that subsequently reported by FRIEDEL and ORCHIN (37). The amount isolated was estimated to be 3.5 mg.

2) Naphthalene (CAS No. 91-20-3): Chromatography (alumina) of the naphthalene-rich fraction yielded 28 mg of crystalline naphthalene, m.p. 78–80 °C. The m.m.p. with an authentic sample, m.p. 80.5–81.0 °C, gave no depression. Its UV absorption spectrum maxima were at 257, 267, 277, 287, and 311 mµ. The UV and IR spectra were identical with those of authentic naphthalene. Treatment with TNF yielded the naphthalene:TNF complex, m.p. 151–155 °C [reported 151.2–154.0 °C (22)]. The m.m.p. with an authentic sample prepared as described by ORCHIN *et al.* (22) showed no depression. The IR spectra of the TNF:isolate complex and TNF:naphthalene complex were identical.

3) 1-Methylnaphthalene (CAS No. 90-12-0): The 1methylnaphthalene-rich fractions on chromatography (alumina) yielded an additional 4 mg of crystalline naphthalene (total = 32 mg) and 3.8 mg of the colorless oil 1-methylnaphthalene. The UV spectrum of the isolate (maxima at 271, 282, 292, 313 mµ) was identical with that of an authentic sample. Treatment with TNF yielded the 1methylnaphthalene:TNF complex, m.p. 161–162 °C [reported 163.4–164.8 °C (38, 39)]. M.m.p. with an authentic sample (38, 39) gave no depression.

4) 2-Methylnaphthalene (CAS No. 91-57-6): The 2methylnaphthalene-rich fractions on chromatography (alumina) yielded an additional 1.2 mg of 1-methylnaphthalene (total = 5.0 mg) and 14 mg of 2-methylnaphthalene as colorless oils. The UV spectrum of the latter isolate was identical with that of an authentic sample. Treatment with TNF yielded the orange crystalline 2methylnaphthalene:TNF complex, m.p. 122-124 °C [reported 124.6-126.0 °C (22]. M.m.p. with an authentic sample (22) gave no depression. The 2-methylnaphthalene:TNF complex was a 1:2 complex as described earlier (38, 39). The quantities of naphthalene, 1-methyl- and 2methylnaphthalene isolated from the CSC were minimal values since these PAHs were particularly prone to sublimation (naphthalene) or distillation (the methyl derivatives).

5) 2-Phenylnaphthalene (CAS No. 612-94-2): Examination of the UV absorption spectra of several fractions following those containing phenanthrene indicated the presence of a PAH suspected to be 2-phenylphenanthrene because of the absorption maxima at 250 and 287 m $\mu$  (37).

6) Acenaphthylene, 1,2-dihydro- (acenaphthene) (CAS No. 83-32-9): Chromatography (alumina) of the acenaphthylene- and 1,2-dihydroacenaphthylene-rich fractions yielded 24 mg of crystalline 1,2-dihydroacenaphthylene, m.p. 94.0–95.0 °C. M.m.p. with an authentic sample gave no depression. Treatment with TNF yielded bright red needles, m.p. 169–175 °C. Recrystallization raised the m.p. to 174.5–176.0 °C. M.m.p. with authentic 1,2-dihydroacenaphthylene:TNF [m.p. 175 °C (22)] gave no depression. Treatment of the isolated PAH with picric acid

<sup>&</sup>lt;sup>c</sup> To conduct the chromatography, this 6.0-m column (assembled from three shorter Pyrex tubes by SHAFFNER) was set up in a stair well and carefully packed with the alumina.

<sup>&</sup>lt;sup>d</sup>In RODGMAN and COOK (2), 44 Figures are depicted in which miniaturized versions of the UV absorption spectra of the various PAH isolates are compared with those of authentic PAH samples plus miniaturized versions of the UV absorption spectra of the several diones are compared to those of authentic samples.

#### Table 3. Column chromatography of the polycyclic aromatic hydrocarbon mixture

Fraction No.	Eluting solvent	Fractions combined	Components
1-75	hexane	2-3 4-7 8-13 14-18 19-22 23-30 31-34 35-41 42-52 53-60 61-75	naphthalene; 1,2-dihydro-4-methylnaphthalene; 1-methylnaphthalene; 2-methylnaphthalene; 2-phenylnaphthalene; acenaphthylene; 1,2-dihydroacenaphthylene; 9 <i>H</i> -fluorene; phenanthrene; dimethylphenanthrene; 9-methylphenanthrene
76-85 86-95 96-105 106-115	2% benzene-hexane 5% benzene-hexane 8% benzene-hexane 10% benzene-hexane	76-84 85-90 91-96 97-105 106-117	anthracene; 9-methylanthracene; pyrene; 1-methylpyrene; 2-methylpyrene; 4-methylpyrene; fluoranthene
116-125 126-135	13% benzene-hexane 16% benzene-hexane	118-126 127-134	fluoranthene, methylfluoranthene
136-145 146-165	20% benzene-hexane 25% benzene-hexane	135-145 146-150 151-158 159-153	benzo[c]phenanthrene; benzo[ <i>gh</i> ]fluoranthene; benzo[/]fluoranthene; benzo[ <i>k</i> ]fluoranthene; benz[e]acephenanthrylene
166-245	35% benzene-hexane	164-172 173-179 180-193 194-197 198-201 201-210 211-231 232-246	benz[a]anthracene; 5-methylbenz[a]anthracene; 2,3-dihydro-1 <i>H</i> -benzo[a]cyclopent[ <i>h</i> ]anthracene; 10,11-dihydro-9 <i>H</i> -benzo[a]cyclopent[ <i>i</i> ]anthracene; chrysene; methylchrysenes; perylene; benzo[a]pyrene; methylbenzo[a]pyrene; benzo[e]pyrene; benzo[ <i>ghi</i> ]perylene
246-255 256-265 266-285 286-295	45% benzene-hexane 50% benzene-hexane 60% benzene-hexane 80% benzene-hexane	247-252 253-272 273-288 289-295	benzo[ <i>ghi</i> ]perylene; benzocyclopentanthracene; dibenz[ <i>a</i> , <i>h</i> ]anthracene
296-305	benzene	296-303- 304-305	dibenz[a,h]anthracene
306-315 316-325	ethyl acetate methanol	306-311 312-318 319-321 322-325	dibenzo[ <i>de,qr</i> ]naphthacene; naphtho[2,1,8- <i>qra</i> ]naphthacene dibenzo[ <i>b,def</i> ]chrysene; dibenzo[ <i>def,mno</i> ]chrysene; dibenz[ <i>a,e</i> ]aceanthrylene; benzo[ <i>rst</i> ]pentaphene; coronene

in 1:1 ethanol:water, followed by crystallization from benzene, yielded 1,2-dihydroacenaphthylene:picric acid complex as orange needles, m.p. 159–160 °C [reported m.p. 160-161 °C (40)]. The UV (maxima at 279, 290, 300, 306, 321 mµ) and IR spectra of the isolate and the IR spectrum of the 1,2-dihydroacenaphthylene:TNF complex were identical with those of authentic samples of the PAH and its TNF complex.

7) Acenaphthylene (CAS No. 208-96-8): The acenaphthylene-rich fractions from the chromatograms described above were combined and rechromatographed (alumina) to yield 7.0 mg of colorless, crystalline hydrocarbon, m.p. 90–91 °C, which on recrystallization melted at 91.5-92.0 °C. The UV (maxima at 266, 324, 339 m $\mu$ ) and IR spectra of the isolate were identical with those of authentic acenaphthylene, m.p.

91-92 °C, prepared by lead oxide oxidation of 1,2dihydroacenaphthylene. M.m.p. of the isolate with authentic acenaphthylene gave no depression.

8) 9H-Fluorene (CAS No. 86-73-7): The 9H-fluorene-rich fractions were successively chromatographed (alumina) three times to yield 3.2 mg of a colorless, crystalline solid, m.p. 113.5-114.5 °C. Its UV spectrum (maxima at 263, 289, 300 mµ) was identical with that of authentic 9H-fluorene. M.m.p. of the isolate and authentic acenaphthylene, m.p. 115.5-116.5 °C, gave no depression. Treatment with TNF yielded a crystalline isolate: TNF complex, m.p. 176-177 °C. M.m.p. of the isolate:TNF complex with authentic 9Hfluorene: TNF complex, m.p. 179.0-179.5 °C, prepared by the method of ORCHIN et al. (22), gave no depression, m.p. 177.5-179.0 °C.

9) Phenanthrene (CAS No. 85-01-8): The phenanthrenerich fractions were chromatographed (alumina) to yield a colorless, crystalline solid, m.p. 97–98 °C. UV (maxima at 252, 274, 281, 293 m $\mu$ ) and IR absorption studies indicated the presence of 1.95 mg of PAH with spectra identical with that of authentic phenanthrene. A m.m.p. study with authentic phenanthrene, m.p. 99.0–99.5 °C, gave a m.p. 98–99 °C. Treatment of the isolate with TNF gave a complex, m.p. 194–196 °C, which gave no depression on admixture with an authentic sample, m.p. 196–197 °C, of phenanthrene:TNF prepared by the method of ORCHIN *et al.* (22).

10) 9-Methylphenanthrene (CAS No. 883-20-5): Fractions following the phenanthrene fractions had UV absorption spectra very similar to that of 9-methylphenanthrene (maxima at 256, 278, 285, 297 m $\mu$ ). The yield was estimated as 0.6 mg.

11) Dimethylphenanthrene (CAS No. 29062-98-4): Fractions following the phenanthrene and 9-methyl homolog fractions appeared to contain other phenanthrene homologs. Their UV absorption spectra were similar to those of dimethylphenanthrenes but not all of the maxima or minima matched those of the 1,2-, 1,3-, 1,7-, 1,8-, 1,9-, 2,3-, 2,5-, 4,5-, or 4,9-dimethylphenanthrenes. The yield was estimated as 2.0 mg.

12) Anthracene (CAS No. 120-12-7): The anthracene-rich fractions were repeatedly chromatographed (alumina) to yield 3.85 mg of crystalline anthracene, m.p. 210–212 °C. M.m.p. with authentic anthracene, m.p. 215–216 °C, gave no depression. Treatment with TNF gave anthracene: TNF complex, m.p. 190–193 °C, as deep red needles. Crystallization from ethanol raised the m.p. to 193.0–194.5 °C [m.p. reported as 193.8–194.0 °C (22)]. M.m.p. with authentic anthracene: TNF complex, m.p. 193.1–193.8 °C, prepared as described by ORCHIN *et al.* (22), gave no depression. The UV (maxima at 252, 309, 323, 338, 356, 376 mµ) and IR absorption spectra were identical with those of authentic anthracene.

13) 9-Methylanthracene (CAS No. 779-02-2): Chromatography (alumina) of the fractions following the anthracene fractions gave fractions with UV absorption (maxima at 302, 315, 332, 348, 367, 387 m $\mu$ ) identical with that of 9methylanthracene. Several of the later factions in the sequence indicated contamination with traces of pyrene as indicated by UV absorption at 333 m $\mu$ .

14) Pyrene (CAS No. 129-00-0): Repeated chromatography (alumina) of the pyrene-rich fractions yielded 0.85 mg of a colorless, crystalline solid, m.p. 148–150 °C (reported 149–150 °C) with UV absorption (maxima at 230, 241, 251, 263, 272, 292, 305, 318, 333, 352, 356, 362, 372 mµ) identical with that of authentic pyrene. Treatment of the isolate with TNF yielded pyrene:TNF complex as red needles, m.p. 239–241 °C. M.m.p. with an authentic pyrene:TNF complex sample (22), m.p. 241.5–242.5 °C, gave no depression.

15) 1-Methylpyrene (CAS No. 2381-21-7): The 1-, 3-, and 4-methylpyrenes were obtained by repeated chromatography (alumina) of the fractions following the pyrene fractions. The UV absorption (272, 308, 321, 337 m $\mu$ ) for one isolate was identical with that of authentic 1-methylpyrene, weight 0.85 mg.

16) 2-Methylpyrene (CAS No. 3442-78-2): UV absorption

maxima at 294, 313, 328, 343, 353 m $\mu$ ) indicated 50  $\mu$ g of 2-methylpyrene. Several other chromatographic fractions (alumina) indicated a mixture of 1- or 4-methylpyrene and 2-methylpyrene but repeated column chromatography did not result in further separation.

17) 4-Methylpyrene (CAS No. 3353-12-6): As indicated by agreement of the UV absorption spectrum (maxima at 273, 296, 307, 318, 336 mµ) of the isolate with that of authentic 4-methylpyrene, 0.72 mg of 4-methylpyrene was isolated. 18) Fluoranthene (CAS No. 206-44-0): Repeated chromatography (alumina) of the fluoranthene-rich fractions yielded crystalline fluoranthene, m.p. 106.5-108.5 °C (reported 110.0-110.5 °C). M.m.p. with authentic fluoranthene gave no depression. UV absorption studies (absolute ethanol, maxima at 245, 253, 262, 272, 276, 283, 288, 309, 323, 342, 359 mµ) indicated 3.45 mg. The IR absorption of the isolate was identical with that of authentic fluoranthene. Treatment with TNF yielded the orange crystalline fluoranthene:TNF complex, m.p.. 205–207 °C. Crystallization from ethanol raised the m.p. to 210–212 °C (reported 215–216 °C). M.m.p. with authentic fluoranthene :TNF complex (22), m.p. 215–216 °C, gave no depression.

19) Methylfluoranthene (CAS No. 30997-39-8): Methylfluoranthene, possibly more than one, was isolated from the fractions following the fluoranthene fractions. Identification was based on the similarities between UV absorption spectra of the isolate and that of fluoranthene and the shift of the maxima to slightly longer wavelengths. Based on UV absorption, the amount of methylfluoranthene was estimated at 0.60 mg.

20) Benzo[*c*]phenanthrene (CAS No. 195-19-7): Trace amounts (approximately 10  $\mu$ g) of spectrally pure benzo[*c*]phenanthrene were obtained chromatographically (alumina). UV absorption maxima were observed at 280, 314, 327, 254, and 371 m $\mu$  in agreement with those of an authentic sample. Several other fractions appeared to contain additional traces of benzo[*c*]phenanthrene but no attempt was made to estimate the amount.

21) Benzo[*ghi*]fluoranthene (CAS No. 203-12-3): A very small quantity (approximately 30  $\mu$ g) of spectrally pure benzo[*ghi*]fluoranthene was obtained with UV absorption maxima at 280, 290, 321, 349, 381, and 402 mµ.

22) Benzo[*j*]fluoranthene (CAS No. 205-82-3): Trace amounts (approximately 10  $\mu$ g) of benzo[*j*]fluoranthene were obtained by chromatography (alumina). UV absorption maxima were noted at 282, 293, 308, 317, 331, 347, 364, 375, and 383 mµ, identical with those of an authentic sample.

23) Benzo[k]fluoranthene (CAS No. 207-08-9): Trace amounts (approximately 10  $\mu$ g) of benzo[k]fluoranthene were obtained by chromatography (alumina). UV absorption maxima were recorded at 296, 308, 358, 378, and 401 m $\mu$ . The following problem is often encountered in a situation involving separation of components of a subfraction of a complex mixture: Several fractions were obtained with UV absorption maxima indicative of possible mixtures of benz[e]acephenanthrylene, benzo[j]fluoranthene, benzo[k]fluoranthene, and/or benzo[ghi]fluoranthene but no attempt was made to estimate the amounts of these PAHs in the fractions. Several attempts to separate them chromatographically (alumina) were unsuccessful. 24) Benz[*e*]acephenanthrylene (CAS No. 205-99-2): Benz[*e*]acephenanthrylene (estimated amount, 0.63 mg) was obtained by chromatography (alumina). UV absorption maxima at 240, 257. 278, 291, 295, 302, 340, 352, and 370 m $\mu$  agreed with those of an authentic sample.

25) Benz[*a*]anthracene (B[*a*]A) (CAS No. 56-55-3): B[*a*]Arich fractions were repeatedly chromatographed (alumina) to yield an isolate with UV absorption maxima at 254, 267, 280, 290, 299, 316, 329, 344, 359, 366, 375, and 384 mµ. The maxima were identical with those of an authentic sample of B[*a*]A. Its amount was estimated at 227 µg. The sample size did not permit too precise a m.p. determination, softening at 150 °C, m.p. 154–155 °C. Treatment with TNF gave a red solid, m.p. 220–222 °C [reported 223–224 °C for B[*a*]A:TNF (22)]. Decomposition of the B[*a*]A:TNF complex on alumina, oxidation of the PAH with 3.0 mg of chromic acid in 1.0 mL of acetic acid, dilution with water (20 mL), followed by extraction with benzene (5 mL) yielded a compound whose UV absorption spectrum was identical with that of B[*a*]A-7,12-dione.

26) 5-Methylbenz[*a*]anthracene (CAS No. 2319-96-2): A PAH with a UV absorption spectrum essentially the same as that of 5-methylbenz[*a*]anthracene was isolated (maxima at 280, 291, 300, 327, 340, 357, 367, 388 m $\mu$ ). Its amount was estimated at 50  $\mu$ g.

27) 2,3-Dihydro-1*H*-benzo[*a*]cyclopent[*h*]anthracene (CAS No. 7099-43-6) and 28). 10,11-Dihydro-9*H*-benzo[*a*]cyclopent[*i*]anthracene (CAS No. 7099-42-5): Fractions contained isolates with UV absorption maxima which agreed with those of 2,3-dihydro-1*H*-benzo[*a*]cyclopent[*h*]anthracene (282, 293, 321, 336, 352, 370, 388 mµ) and 10,11-dihydro-9*H*-benzo[*a*]cyclopent[*i*]anthracene (280, 297, 302, 316, 320, 438, 360, 387 mµ) (42). Trace amounts (about 10 µg) of each of these PAHs were obtained but some background UV absorption shifted the maxima slightly to longer wavelengths but not enough for the change to be interpreted as due to the presence of a methyl group.

29) Chrysene (CAS No. 218-01-9): Repeated chromatography (alumina) of the chrysene-rich fractions yielded several fractions with absorption maxima at 268, 284, 295, 307, 319, 344, 353, and 363 mµ, in agreement with those observed for authentic chrysene. The crystalline material melted at 249–251 °C. Treatment wit TNF yielded the chrysene:TNF complex, yellow needles, m.p. 246–248 °C [reported m.p. 247.8–249.0 °C (22)]. It was estimated that 0.62 mg of chrysene was present. Several other fractions obviously contained chrysene admixed with chrysene-like compounds, possibly one or more methylchrysenes.

30). Methylchrysenes (CAS No. 41637-90-5): Rechromatography of the chrysene-methylchrysenes fraction yielded approximately 0.3 mg of methylchrysenes with UV absorption maxima at 269, 297, 313, 328, 348, and 366 mµ. These UV maxima did not exactly match those recorded for 1-methylchrysene, 4-methylchrysene, 5-methylchrysene, or 6-methylchrysene. A mixture of them was proposed.

31) Perylene (CAS No. 198-55-0): Repeated chromatography (alumina) of the perylene-rich fractions yielded approximately 0.2 mg of spectrally pure perylene with UV absorption maxima (244, 252, 263, 343, 367, 388, 408, 434 m $\mu$ ) that agreed with those of an authentic sample. It was estimated that an additional 0.1 mg of perylene was present in fractions preceding and subsequent to those fractions

containing the spectrally pure perylene.

32) Benzo[*a*]pyrene (B[*a*]P) (CAS No. 50-32-8): The B[*a*]Prich fractions were chromatographed (alumina) repeatedly to yield 0.905 mg of B[*a*]P, m.p. 178–180 °C, m.m.p. with authentic B[*a*]P, m.p. 180–181 °C, gave no depression. The UV absorption spectrum (maxima at 267, 273, 286, 297, 333, 348, 364, 380, 386, 403 mµ) was identical at each maximum and each minimum with the UV absorption spectrum of an authentic sample of B[*a*]P. M.m.p. of the isolate:picric acid and authentic B[*a*]P:picric acid complexes, prepared as described by FIESER (41), gave no depression.

33) Methylbenzo[*a*]pyrene (CAS No. 25167-89-9): A sample (80  $\mu$ g) of a PAH was obtained by repeated chromatography (alumina) with a UV absorption spectrum almost identical with that of B[*a*]P, except each maximum was shifted to a slightly longer wavelength. This suggested a methyl-B[*a*]P. 34) Benzo[*e*]pyrene (B[*e*]P) (CAS No. 192-97-2): Repeated chromatography (alumina) of the B[*e*]P-rich fractions yielded 0.62 mg of B[*e*]P with UV absorption maxima (278, 289, 304, 318, 322, 348, 358, 367 mµ) identical with those of an authentic sample.

35) Benzo[ghi]perylene (CAS No. 191-24-2): BY repeated chromatography (alumina), benzo[ghi]perylene (0.24 mg) was obtained with UV absorption maxima at 291, 303, 331, 348, 367, 388, 406 mµ. UV absorption spectra indicated that several minor isolated fractions were mixtures of B[a]P and benzo[ghi]perylene.

36) Benzocyclopentanthracene: A small quantity (90  $\mu$ g) of a benzocyclopentanthracene was obtained by repeated chromatography (alumina) of benzocyclopentanthracene-rich fractions. UV absorption maxima were obtained at 284, 296, 328, 342, 358, 376, and 390 m $\mu$ , most of which agreed with those of an authentic sample of the PAH 1,2dihydrobenz[*j*]aceanthrylene (cholanthrene). Attempts to prepare a PAH:TNF complex suitable for a m.p. determination were unsuccessful because of the sample size. However, an olive-green color characteristic of 1,2dihydrobenz[*j*]aceanthrylene:TNF complex was obtained by treatment of the PAH with one equivalent of TNF in an ethanol-benzene solution. As noted previously, the absence of this PAH from the many PAHs identified by SNOOK *et al.* (19) indicated that our original assignment was erroneous.

37) Dibenz[a,h]anthracene (DB[a,h]A) (CAS No. 53-70-3): The DB[a,h]A-rich fractions were repeatedly chromatographed (alumina) with hexane-benzene solutions as eluents. The amount of DB[a,h]A was estimated at 0.205 mg, based on calculation from its UV absorption spectrum. Treatment of the DB[a,h]A with 3 mg of chromic oxide in 1.5 mL of glacial acetic acid yielded 0.225 mg of DB[a,h]A-7,12-dione, based on calculation from its UV absorption spectrum. Thus the oxidation gave a 90% yield. UV absorption maxima (230, 278, 290, 301, 323, 337, 351, 373, 386, 396 mµ) for the DB[a,h]A isolate and absorption maxima (297, 334, 390 mµ) for the DB[a,h]A-7,12-dione were determined in absolute ethanol.

These agreed with UV absorption maxima obtained with authentic sample of DB[a,h]A and DB[a,h]A-7,12-dione, respectively. The IR absorption spectrum of the dione, while not too definitive, did agree substantially with that of an authentic sample.

38) Dibenzo[*de*,*qr*]naphthacene (CAS No. 193-09-9) and 39) Naphtho[2,1,8-*qra*]naphthacene (CAS No. 196-42-9): The reported UV absorption maxima for dibenzo[*de,qr*]naphthacene were 297, 309, 321, 328, 345, 367, 375, 386, 396, and 408 mµ whereas those reported for naphtho[2,1,8-*qra*]naphthacene were 247, 286, 297, 320, 335, 385, 406, and 458 mµ (42). A PAH sample was obtained for which all of the above-listed maxima were present (247, 286, 297, 309, 320, 328, 335, 345, 367, 375, 385, 396, 407, 430, 458 mµ). It was concluded that both PAHs were present in the sample. 40) Dibenzo[*b,def*]chrysene (CAS No. 189-64-0): Repeated chromatography (alumina) yielded a trace amount (about 10 µg) of dibenzo[*b,def*]chrysene with UV absorption maxima (290, 302, 314, 379, 401, 425, 452 mµ) identical with those of an authentic sample

41) Dibenzo[*def,mno*]chrysene (CAS No. 191-26-4): About 100  $\mu$ g of spectrally pure dibenzo[*def,mno*]chrysene were obtained by repeated chromatography (alumina) of the dibenzo[*def,mno*]chrysene-rich fractions. UV absorption maxima (260, 297, 309, 385, 408, 437 m $\mu$ ) of the isolate agreed with those obtained with an authentic sample. Some overlapping was noted between several fractions containing dibenzo[*def,mno*]chrysene and other PAHs of unknown structure.

42) Dibenz[*a*,*e*]aceanthrylene<sup>e</sup> (CAS No. 5385-75-1): Repeated chromatography (alumina, then Fluorisil®) yielded a trace amount (about 20  $\mu$ g) of spectrally pure dibenz[*a*,*e*]aceanthrylene (dibenz[*a*,*e*]fluoranthene) with UV absorption maxima at 291, 305, 319, 332, 352, 363, 374, 403, and 438 mµ.

43) Benzo[*rst*]pentaphene (CAS No. 189-55-9): Repeated chromatography (alumina) yielded a trace amount (about 20  $\mu$ g) of dibenzo[*rst*]pentaphene with UV absorption maxima (274, 284, 297, 316, 330, 354, 375, 395, 419 m $\mu$ ) identical with those of an authentic sample. UV absorption analyses of several other fractions indicated they contained mixtures of dibenzo[*b*,*def*]chrysene, benzo[*rst*]pentaphene, and dibenzo[*def*,*mno*]chrysene but no clear-cut separation could be effected by repeated column chromatography.

44) Coronene (CAS No. 191-07-1): Approximately 50  $\mu$ g of coronene were isolated. Its UV absorption spectrum showed maxima at 328, 338, 342 and 388 m $\mu$ .

UV absorption spectra devoid of background absorption were obtained for many fractions obtained from the lengthy chromatogram. Even though these showed spectral characteristics typical of PAHs, no structural assignments could be made for any of the suspected PAHs despite diligent examination of much UV absorption data catalogued in the literature by FRIEDEL and ORCHIN (37), CLAR (42), and others. It was suspected that many mixtures of PAHs were obtained and that such mixtures contained PAHs already identified in this study plus PAHs that could not be identified.

#### DISCUSSION

The fact that over 40 PAHs were identified in this study and about 50 were identified in other pre-1964 studies by more or

less classical chemical means indicates the limitations of the technologies involved in the so-called classical chemical technology. The first PAH identified in tobacco smoke was azulene in 1947 by IKEDA (43). However, earlier in the late 1930s-early 1940s, ROFFO and his son had reported the identification of anthracene, phenanthrene, and B[a]P in the destructive distillate of tobacco (44, 45). Their findings were much discounted by various authorities who insisted that a destructive distillate of tobacco had no relationship to tobacco smoke [see p. 259 in WYNDER and HOFFMANN (12), p. 94 in WYNDER and HOFFMANN (13), pp. 422-424 in LARSON et al. (46)]. Many of the findings reported by ROFFO on the chemical and biological properties of the destructive distillate of tobacco and on the destructive distillate of organic solvent-extracted tobacco parallel the qualitative but not quantitative properties subsequently reported on CSC from control and organic solvent-extracted tobaccos (47).

Several situations subsequently stimulated substantial interest in the possibility of the presence of PAHs in cigarette smoke:

- ► The publication in the early 1950s of the results obtained in a series of epidemiological studies, each of which indicated the adverse effect of cigarette smoking on the incidence of lung cancer in cigarette smokers (48).
- ► The 1953 report by WYNDER *et al.* (23) of the induction of skin carcinomas in a significant percentage of laboratory animals (mice) skin painted repeatedly for many months with concentrated solutions of cigarette "tar."
- The number of reports listed by HARTWELL in his 1951 survey of tumorigenic compounds (25) on tumor induction in laboratory animals skin painted or injected subcutaneously with solutions of various individual PAHs. The biological results obtained with more than 270 PAHs were summarized by HARTWELL. Examination of his tabulation reveals that 99 of the PAHs and their alkyl derivatives tested (36.5%) exhibited tumorigenicity. Very few bicyclic or tricyclic PAHs were reported as tumorigenic. For the total of 232 tetracyclic, pentacyclic, and hexacyclic PAHs and their alkyl derivatives listed by HARTWELL (25), the number reported as tumorigens was 96 (41.4%).

As a result of the increased effort to study PAHs in cigarette smoke, the number of identified PAHs in cigarette smoke had increased to almost 100 by late 1963. Over the next decade and a half with the incredible advancements in fractionation and analytical technologies and their utilization in the characterization of the components of complex mixtures such as cigarette smoke, the number of identified PAHs in cigarette smoke was increased more than five-fold from the fewer than 100 to more than 500 [see research outline by SNOOK et al. (19), PAH catalogue by RODGMAN and PERFETTI (20)]. An additional factor is that many of the later characterizations by SNOOK et al. (19) were accomplished by smoking far fewer cigarettes than the 20000 cigarettes smoked in the RODGMAN-COOK study (3) to obtain sufficient CSC for fractionation and study of the PAH-rich fractions by classical chemical means. While a plethora of different types of spectral and graphical data (UV, IR, NMR, and mass spectra plus gas chromatograms) have been used to identify literally

<sup>&</sup>lt;sup>e</sup> For many years, dibenz[*a,e*]aceanthrylene (also known occasionally as dibenzo[*a,e*]fluoranthene) was catalogued as dibenzo[*def,p*]chrysene (known chronologically as 1,2,3,4-dibenzopyrene or 3,4,8,9dibenzopyrene and then dibenzo[*a,l*]pyrene) but a 1966 study by LAVIT-LAMY and BUU-HOÏ (33) defined the compound originally thought to be dibenzo[*a,l*]pyrene as its isomer dibenz[*a,e*]aceanthrylene.

hundreds of the PAHs in cigarette smoke, the number isolated in crystalline form or as oils has remained relatively few, totaling less than 3% of the more than 500 PAHs identified in cigarette smoke.

The more recent use of advanced analytical technologies and the generation of much meaningful spectral and graphical data rather than the earlier classical chemical methods as the basis for the identification of literally hundreds of the chemical components in tobacco and tobacco smoke is not being criticized. Examination of the procedures used by SNOOK *et al.* in their studies of PAHs (19) and aza-arenes (49) in tobacco smoke, those used by NEWELL *et al.* (50), SCHUMACHER *et al.* (51), and HECKMAN and BEST (52) in their studies on cigarette smoke composition <sup>f</sup>, and those used by LLOYD *et al.* (53), LEFFINGWELL and ALFORD (54), and PENG *et al.* (55) in their studies on tobacco composition indicates the great number of chemical components unequivocally identified as well as the paucity of identification errors.

Table 4 summarizes many of the references on the chronological identification of the 97 PAHs identified in tobacco smoke prior to the 1964 publication of the famous Advisory Committee's report to the U.S. Surgeon General (15). Of the 97 PAHs listed, 91 had been described in nearly 200 pre-1964 publications. Surprisingly, in the chapter on tobacco and tobacco smoke composition [Chapter 6 in (15)], the Advisory Committee member (FIESER) and his consultant (ORCHIN) who authored the chapter discussed in detail 18 named PAHs. Of the 18, 13 named PAHs<sup>g</sup> were described as cigarette smoke components (acenaphthylene, 1,2-dihydroacenaphthylene, anthracene, B[a]A, benzo[*i*]fluoranthene, benzo[*rst*]-pentaphene, benzo[ghi]pervlene, benzo[c]phenanthrene, B[a]P, DB[a,h]A, dibenzo[def,p]chrysene, fluorene, phenanthrene, pyrene) but the five other named PAHs were discussed as being present as components of carbon black (benzo[ghi]perylene, benzo[e]pyrene, coronene, dibenzo[def,mno]chrysene, fluoranthene). Several of the 13 named PAHs had reported tumorigenic properties. Although none was mentioned by name, 27 other PAHs that had been identified in cigarette smoke were described as follows:

However, no less than 27 hydrocarbons containing four or more condensed rings which have been tested for carcinogenicity with negative results have been isolated from tobacco smoke tar.

Was the main reason for not naming any of the 27 PAHs specifically because none of the 27 possessed tumorigenic properties or was it because several were known to be effective antitumorigens vs. the very potent tumorigens B[a]P and DB[a,h]A? Several reviewers (LEITER, SELIGMAN) of the draft of the chapter on tobacco and tobacco smoke composition were either complimentary or satisfied with it. However, in a late 1963 letter to the Advisory Committee Staff Director, KOTIN (56) listed the serious shortcomings that he and FALK felt were present in the draft of the chapter. In an earlier article, KOTIN and

FALK (57) had summarized research findings on the anticarcinogenicity vs. B[a]P or vs. DB[a,h]A of nine PAHs (anthracene, benzo[a]fluorene, B[a]A, chrysene, pyrene, B[e]P, benzo[k]fluoranthene, benzo[ghi]fluoranthene, perylene), all of which had been identified in cigarette MSS prior to the Advisory Committee 1964 report and five of which were listed in its review [see Chapter 6 in (15)]. Twice in the Report [see pp. 143 and 144 (15)], anticarcinogens in tobacco smoke were mentioned by citation of a special KOTIN-FALK report to the Committee [see Reference 107, p. 242 in (15)], but no specific anticarcinogen was defined. Of the nearly 200 available publications by year-end 1963 on PAHs in cigarette smoke, only 15 publications were cited by FIESER and ORCHIN, the authors of the tobacco and smoke composition chapter in the Advisory Committee's report.

## ACKNOWLEDGMENTS

The authors wish to acknowledge their great appreciation of the exceptional skill demonstrated by the late John J. Whalen in obtaining the meaningful IR spectra of extremely small samples of various PAHs and their derivatives and subsequently recovering the samples for additional study. Also greatly appreciated is the significant technical assistance provided during this study by two exceptionally talented technicians, Max A. Wagoner and the late Bruce W. Woosley.

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<sup>&</sup>lt;sup>f</sup> In these three smoke composition studies, 1545 components were identified, 825 of which had not been reported previously in cigarette smoke, 720 of which were confirmation of previously reported tobacco smoke components.

<sup>&</sup>lt;sup>g</sup> The nomenclature used in the listing is not that used in or before 1964 but is the current one defined by International Union of Pure and Applied Chemistry (IUPAC).

Table 4.	The polycyclic aromatic	hydrocarbons re	ported in tobacco	smoke prior to 1964 <sup>a</sup> :
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No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
1	Acenaphthylene	208-96-8	<ul> <li>1956 Bonnet and Neukomm (58), Gilbert and Lindsey (59, 60)</li> <li>1957 Bonnet and Neukomm (28)</li> <li>1959 Bonnet and Neukomm (61, 62), Johnstone and Plimmer (63), Pietzsch (64), Zapior <i>et al.</i> (65)</li> </ul>
			1960 Rodgman and Cook (3)
			1962 Lyons (66)
			1963 Cook <i>et al.</i> (67), Philip Morris (16) 1964 Elmenhorst and Reckzeh (18), USPHS (15)
2	Acenaphthylene, 1,2-dihydro-	83-32-9	1953 Cooper and Lindsey (68)
			1954 Cooper and Lindsey (69), Cooper et al. (70)
			1955 Cooper and Lindsey (71), Lindsey (72, Lyons (73))
			1956 Bonnet and Neukomm (58), Campbell and Lindsey (74, Gilbert and Lindsey (59, 60), Rodgman (1)
			1957 Bonnet and Neukomm (28), Campbell and Lindsey (75), Wynder and Wright (76)
			1958 Wynder <i>et al.</i> (77)
			1959 Bonnet and Neukomm (61, 62), Johnstone and Plimmer (63), Lindsey (78), Pietzsch (64)
			1960 Rodgman and Cook (3)
			1961 Alexandrov <i>et al.</i> (79) 1962 Lyons (66)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15)
3	Anthracene	120-12-7	1953 Cooper and Lindsey (68)
			1954 Commins <i>et al.</i> (80), Cooper and Lindsey (69), Cooper <i>et al.</i> (70)
			1955 Cooper and Lindsey (71), Lindsey (72), Lyons (73), Seelkopf (81)
			1956 Bonnet and Neukomm (58), Campbell and Lindsey (74), Gilbert and Lindsey (59, 60), Rodgman (1), Wright and Wynder (82)
			1957 Bonnet and Neukomm (28), Campbell and Lindsey (75), Wynder and Wright (76)
			1958 Ahlmann (83), Pietzsch (84), Scassellati-Sforzolini and Salucci (85), Wynder and Wright (77)
			1959 Bonnet and Neukomm (62), Johnstone and Plimmer (63), Lindsey (78, 86), Pietzsch (64), Zapior <i>et al.</i> (65)
			1960 Candeli <i>et al.</i> (87), Cuzin (88), Mouron <i>et al.</i> (89), Rodgman and Cook (3)
			1960/61 Wynder (90) 1961 Sasmoco (91), Scassellati-Sforzolini and Mariani (92, 93), Scassellati-Sforzolini and Saldi (94)
			1962 Lyons (66), Parmele and Jensen (95), Pietzsch (96)
			1963 Cuzin et al. (97), Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15)
4	Anthracene, alkyl-		1955 Cooper and Lindsey (71), Lyons (73)
5	Anthracene, 9,10-dihydro-	613-31-0	1963 Philip Morris (16) 1959 Pietzsch (64)
5	Antinacene, 3, 10-dinydio-	013-31-0	1964 Elmenhorst and Reckzeh (18)
6	Anthracene, methyl-	26914-18-1	1956 Dickey and Touey (98), Tennessee Eastman (99)
7	Anthracene, 2-methyl-	613-12-7	1955 Cooper and Lindsey (71)
			1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
8	Anthracene, 9-methyl-	779-02-2	1964 Elmenhorst and Reckzeh (18) 1960 Rodgman and Cook (3)
9	Azulene	275-51-4	1947 Ikeda (43)
			1955 Lyons (73)
			1956 Gilbert and Lindsey (59, 60), Lyons (100)
			1958 Lyons (30)
			1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16) 1964 Elmenhorst and Reckzeh (18)
			1964 Elmennorst and Reckter (18)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
11	Benz[/]aceanthrylene, 1,2- dihydro-3- methyl-	56-49-5	1963 Philip Morris (16) 1964 Kröller (101)
12	Benz[e]acephenanthrylene <sup>f</sup>	205-99-2	1957 Wynder and Wright (76)
			1959 Hoffmann and Wynder (102), Johnstone and Plimmer (63), Wynder and Hoffmann (103)
			1960 Hoffmann and Wynder (104, 105)
			1960/61 Wynder (90)
			1961 Wynder and Hoffmann (106)
			1962 Lyons (66)
			1964 Elmenhorst and Reckzeh (18)
13	Pon-Telenthrasens	56-55-3	
13	Benz[a]anthracene	50-55-5	1955 Lettré and Jahn (107), Lyons (73) 1956 Bonnet and Neukomm (58), Gilbert and Lindsey (59, 60).
			Lyons (100), Rodgman (1) 1957 Bonnet and Neukomm (28), Lyons and Johnston (29), Neukomm and Bonnet (108), Pietzsch (64), Rand <i>et al.</i> (109)
			1958 Ahlmann (83), Pietzsch (84), Wynder <i>et al.</i> (77)
			1959 Bonnet and Neukomm (61, 62), Johnstone and Plimmer (63), Pietzsch (64), Wynder and Hoffmann (110)
			1960 Rodgman and Cook (3), Hoffmann and Wynder (105)
			1960/61 Wynder (90)
			1962 Lyons (66), Pietzsch (96)
			1962/63 Pyriki (32)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15)
14	Benz[a]anthracene, 7,12-dimethyl-	57-97-6	1958 Pietzsch (84)
			1959 Johnstone and Plimmer (63), Pietzsch (64)
			1962 Lyons (66)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), Kröller (101)
15	Benz[a]anthracene, 5-methyl-	2319-96-2	1956 Bonnet and Neukomm (58)
			1957 Neukomm and Bonnet (108)
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1962/63 Pyriki (32)
			1963 Philip Morris (16)
16	14 Danzalala valanant (hlanthraaana, 2.2		1964 Elmenhorst and Reckzeh (18),
16	1H-Benzo[a]cyclopent[h]anthracene, 2,3- dihydro-	7099-43-6	1956 Bonnet and Neukomm (58) 1957 Bonnet and Neukomm (28), Lyons and Johnston (29),
			Neukomm and Bonnet (108)
			1958 Bonnet (111), British Empire Cancer Campaign (112)
			1959 Bonnet and Neukomm (61), Johnstone and Plimmer (63) 1960 Rodgman and Cook (3)
			-
			1962/63 Pyriki (32) 1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
17	9H-Benzo[a]cyclopent[i]anthracene, 10,11-	7099-42-5	1956 Bonnet and Neukomm (58)
	dihydro-	1099-42-5	1957 Bonnet and Neukomm (28), Lyons and Johnston (29), Neukomm and Bonnet (108)
			1958 Ahlmann (83), Bonnet (111)
			1959 Bonnet and Neukomm (61), Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1962/63 Pyriki (32)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
18	Benzo[cd]fluoranthene	16135-81-2	1959 Bentley and Berry (113)
		10100 01 2	1963 Philip Morris (16)
19	Benzo[ghi]fluoranthene	203-12-3	1957 Lyons and Johnston (29)
		200 12 0	1958 Van Duuren (114), Van Duuren and Nelson (115), Wynde et al. (77)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
19	Benzo[ghi]fluoranthene (cont.)	203-12-3	1959 Johnstone and Plimmer (63), Wynder and Hoffmann (103, 106)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
			1960 Hoffmann and Wynder (104, 105)
			1960/61 Wynder (90)
			1961 Wynder and Hoffmann (110)
			1964 Elmenhorst and Reckzeh (18)
20	Benzo[/]fluoranthene	205-82-3	1958 Van Duuren and Nelson (115)
			1959 Johnstone and Plimmer (63), Wynder and Hoffmann (103, 106)
			1960 Hoffmann and Wynder (104, 105), Rodgman and Cook (3)
			1960/61 Wynder (90)
			1961 Wynder and Hoffmann (110)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15)
21	Benzo[k]fluoranthene	207-08-9	1956 Lyons (100)
			1957 Lyons and Johnston (29)
			1958 British Empire Cancer Campaign (112), Lyons (30), Van Duuren (114), Van Duuren and Nelson (115)
			1959 Johnstone and Plimmer (63), Tennessee Eastman (), Wynder and Hoffmann (103, 106)
			1960 Rodgman and Cook (3), Hoffmann and Wynder (104, 105)
			1961 Wynder and Hoffmann (110)
			1962 Lyons (66)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
22	5H-Benzo[a]fluorene	238-79-9	1963 Mold <i>et al</i> . (117), Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
23	11 <i>H</i> -Benzo[a]fluorene	238-84-6	1956 Bonnet and Neukomm (58)
			1957 Bonnet and Neukomm (28), Neukomm and Bonnet (108) 1958 Ahlmann (83), Van Duuren and Nelson (115), Wynder <i>et</i> <i>al.</i> (77)
			1962 Pietzsch (96)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
24	11H-Benzo[a]fluorene, 11-methyl-	71265-25-3	1956 Bonnet and Neukomm (58)
			1957 Neukomm and Bonnet (108)
			1958 Ahlmann (83)
			1959 Johnstone and Plimmer (63), Tennessee Eastman (116)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
25	11 <i>H</i> -Benzo[b]fluorene	243-17-4	1958 Ahlmann (83)
		2.0	1959 Johnstone and Plimmer (63)
			1960 Hoffmann and Wynder (105)
			1960/61 Wynder (90)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
26	7H-Benzo[c]fluorene	205-12-9	1954 Einfeinfolst and Reckzen (16) 1958 Wynder <i>et al.</i> (77)
20		200-12-0	1964 Elmenhorst and Reckzeh (18)
27	Benzo[a]naphthacene	226-88-0	1957 Wynder and Wright (76)
	=	220 00 0	1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
			1963 Fillip Moris (16) 1964 Elmenhorst and Reckzeh (18)
28	Benzo[ <i>rst</i> ]pentaphene	189-55-9	1964 Einemoist and Reckzen (18)
20	Perizoli orliferitati i ci i c		1957 Bonnet and Neukomm (28), Neukomm and Bonnet (108), Wynder and Wright (76)
			1958 Bonnet (111), Wynder <i>et al.</i> (77)
			1959 Bonnet and Neukomm (61, 62)
			1960 Rodgman and Cook (3)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
28	Benzo[rsf]pentaphene (cont.)	189-55-9	1962 Neukomm (118), Pietzsch (96) 1962/63 Pyriki (32)
			1963 Gil-Av and Shabtai (119) 1964 Elmenhorst and Reckzeh (18), USPHS (15)
29	Benzo[ <i>ghi</i> ]perylene	191-24-2	1954 Cooper and Lindsey (69)
			1955 Cooper and Lindsey (71), Lindsey (72), Lyons (73) 1956 Rodgman (1)
			1957 Lyons and Johnston (29)
			1958 Ahlmann (83), British Empire Cancer Campaign (112), Pietzsch (84), Van Duuren (120), Van Duuren and Nelson (115), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Pietzsch (64), Tennessee Eastman (116), Wynder and Hoffmann (103, 110)
			1960 Hoffmann and Wynder (104, 105), Rodgman and Cook (3)
			1960/61 Wynder (90)
			1962 Lyons (66)
			1962/1963 Pyriki (32) 1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15) <sup>d</sup>
30	Benzo[c]phenanthrene	195-19-7	1958 British Empire Cancer Campaign (112), Van Duuren (114), Van Duuren and Nelson (115), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
24	Denzelelnurene	E0 22 9	1964 Elmenhorst and Reckzeh (18), USPHS (15)
31	Benzo[a]pyrene	50-32-8	1952 Hofer and Peebles (38) 1954 Cooper and Lindsey (69), Cooper <i>et al.</i> (70)
			1955 Cardon and Alvord (6), Alvord and Cardon (7), Cooper
			and Lindsey (71), Lettré and Jahn (107), Lindsey (72), Seelkopf (81), Wieske (121, Wright and Wynder (122)
			1955/56 Alvord and Cardon (123), Falk and Kotin (124)
			1956 Babin <i>et al.</i> (125),Bonnet and Neukomm (58), Campbell and Lindsey (74), Cardon <i>et al.</i> (126), Gilbert and Lindsey (59, 60), Kuratsune (127), Lam (128), Lettré <i>et al.</i> (129), Lyons (100), Rodgman (1), Wright and Wynder (82),
			Wynder and Wright (130) 1956/57 Wright (131)
			1950/57 Wright (131) 1957 Bonnet and Neukomm (28), Campbell and Lindsey (75),
			Fieser (9), Lyons and Johnston (29), Neukomm (118, 132), Neukomm and Bonnet (108), Rand <i>et al.</i> (109), Wieske (133), Wynder and Wright (76)
			1958 Ahlmann (83), 1958 Bentley and Burgan (134, Cardon (135), Clemo (136), Cuzin (137), Hubert-Habart <i>et al.</i> (138), Latarjet <i>et al.</i> (139), Orris <i>et al.</i> (10), Pietzsch (84), Scassellati-Sforzolini and Salucci (85), Van Duuren (114, 120, 140), Van Duuren and Nelson (115), Wynder <i>et al.</i> (77)
			1959 Bonnet and Neukomm (61, 62), Burgan (141), Dikun (142, Dikun and Chushkin (143), Hoffmann and Wynder (102), Johnstone and Plimmer (63), Lindsey (78, 86), Lindsey et al. (144), Pietzsch (64), Tennessee Eastman (145), Wynder and Hoffmann (103, 110), Wynder et al. (146), Zapior et al. (65)
			1960 Bentley and Burgan (147, 148), Candeli <i>et al.</i> (149), Clemo 150), Cuzin (88), Cuzin <i>et al.</i> (151),Dikun <i>et al.</i> (152), Hoffmann and Wynder (104, 105), Hubert-Habart (153), Mouron <i>et al.</i> (89), Pavlu and Sula (154), Rodgman and Cook (3), Takayama and Oota (155)
			1960/61 Wynder (90) 1961 Alexandrov <i>et al</i> . (79), Grimmer (156), Guérin (157),
			Mouron <i>et al.</i> (79), Grimmer (156), Guerin (157), Scassellati-Sforzolini and Mariani (92, 93), Scassellati- Sforzolini and Saldi (94), Wynder and Hoffmann (106)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
31	Benzo[a]pyrene (cont.)	50-32-8	1962 Barkemeyer (160), Bell (161), Bentley (162), Dikun <i>et al.</i> (163), Lyons (66), Neukomm (118), Parmele and Jensen (95), Pietzsch (96)
			1962/63 Pyriki (32)
			1963 Cuzin <i>et al.</i> (97), Druckrey and Schildbach (164), Gil-Av and Shabtai (119), Hoffmann <i>et al.</i> (165), Philip Morris (16), Scherbak <i>et al.</i> (166), Testa <i>et al.</i> (167), Waltz and Häusermann (168)
			1964 Elmenhorst and Reckzeh (18), Fieser (11), Kröller (101), USPHS (15)
32	Benzo[ <i>a</i> ]pyrene, alkyl-		1958 Wynder <i>et al.</i> (77)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
33	Benzo[a]pyrene, 7,8-dihydro-	17573-23-8	1958 Ahlmann (83)
			1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
	Develop following a discretion d	05407.00.0	1964 Elmenhorst and Reckzeh (18)
34 35	Benzo[ <i>a</i> ]pyrene, dimethyl- Benzo[ <i>a</i> ]pyrene, methyl-	25167-90-2 25167-89-9	1958 Clemo (136) 1958 Van Duuren (120), Van Duuren and Nelson (115)
55	Denzolajpyrene, metriyi-	23107-03-3	1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
36	Benzo[e]pyrene	192-97-2	1955 Lettré and Jahn (107)
			1957 Lyons and Johnston (29)
			1958 Ahlmann (83), British Empire Cancer Campaign (112), Orris <i>et al.</i> (10), Pietzsch (84), Van Duuren (114, 120, 140), Van Duuren and Nelson (115), Wynder <i>et al.</i> (77)
			1959 Hoffmann and Wynder (102), Johnstone and Plimmer (63), Pietzsch (64), Wynder and Hoffmann (103, 110)
			1960 Hoffmann and Wynder (104, 105), Rodgman and Cook (3
			1960/61 Wynder (90)
			1962 Lyons (66), Pietzsch (96)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15) <sup>d</sup>
37	2,2'-Binaphthalene	612-78-2	1961 Izawa (169) 1956 Weight and Weights (82)
38	Chrysene	218-01-9	1956 Wright and Wynder (82) 1957 Lyons and Johnston (29), Rand <i>et al</i> . (109)
			<ul> <li>1957 Eyons and Johnston (29), Rand et al. (195)</li> <li>1958 Ahlmann (83), British Empire Cancer Campaign (112), Pietzsch (84), Van Duuren (120, Van Duuren and Nelson (115), Wynder <i>et al.</i> (77)</li> </ul>
			1959 Hoffmann and Wynder (102), Johnstone and Plimmer (63), Pietzsch (64), Wynder and Hoffmann (103, 110)
			1960 Hoffmann and Wynder (104, 105), Rodgman and Cook (3 1960/61 Wynder (90)
			1962 Lyons (66)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
39	Chrysene, alkyl-		1958 Van Duuren (114), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Wynder and Hoffmann (103 110)
			1960 Hoffmann and Wynder (104, 105), Rodgman and Cook (3 1960/61 Wynder (90)
			1963 Philip Morris (16), Hoffmann and Wynder (104, 105)
			1964 Elmenhorst and Reckzeh (18)
40	Chrysene, dimethyl-	41637-92-7	1958 Ahlmann (83)
			1959 Johnstone and Plimmer (63)
			1960 Hubert-Habart (153)
			1962/63 Pyriki (32)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
41	Chrysene, 1-methyl-	3351-28-8	1958 Ahlmann (83), Van Duuren and Nelson (115)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
42	Coronene	191-07-1	1956 Gilbert and Lindsey (59, 60)
			1958 Pietzsch (84), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Lindsey (86), Pietzsch (64), Tennessee Eastman (116)
			1960 Candeli et al. (87), Rodgman and Cook (3)
			1962 Lyons (66)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15) <sup>d</sup>
43	15H-Cyclopenta[a]phenanthrene	219-07-8	1959 Tennessee Eastman (116)
			1964 Elmenhorst and Reckzeh (18)
44	15H-Cyclopenta[a]phenanthrene, 16,17-	482-66-6	1958 Wynder <i>et al.</i> (77)
	dihydro-		1959 Tennessee Eastman (116)
			1963 Philip Morris (16)
45		000 04 5	1964 Elmenhorst and Reckzeh (18)
45	4H-Cyclopenta[def]phenanthrene	203-64-5	1963 Philip Morris (16)
40		F0 70 0	1964 Kröller (101)
46	Dibenz[a,h]anthracene	53-70-3	1958 Pietzsch (84), Van Duuren (114), Van Duuren and Nelsor (115), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Pietzsch (64), Tennessee Eastman (116), Wynder and Hoffmann (103, 110)
			1960 Hoffmann and Wynder (104, 105)
			1960 Rodgman and Cook (3)
			1960/61 Wynder (90)
			1962 Lyons (66)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), Kröller (101), USPHS (15)
47	Dibenzo[ <i>b,def</i> ]chrysene	189-64-0	1957 Lyons and Johnston (29), Wynder and Wright (76)
			1958 British Empire Cancer Campaign (112), Wynder et al. (77
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1962/63 Pyriki (32)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
48	Dibenzo[def,mno]chrysene	191-26-4	1954 Cooper and Lindsey (69)
			1955 Cooper and Lindsey (71), Lindsey (72)
			1956 Gilbert and Lindsey (59, 60), Rodgman (1)
			1957 Lyons and Johnston (29)
			1958 Pietzsch (84), Scassellati-Sforzolini and Salucci (85)
			1959 Johnstone and Plimmer (63), Lindsey (86), Pietzsch (64), Tennessee Eastman (116)
			1960 Hoffmann and Wynder (105), Rodgman and Cook (3)
			1962 Lyons (66)
			1964 Elmenhorst and Reckzeh (18), USPHS (15) <sup>d</sup>
49	Dibenzo[ <i>def,p</i> ]chrysene <sup>g</sup>	191-30-0	1957 Rodgman (2)
			1958 Lyons (30)
			1959 Johnstone and Plimmer (63), Schoental (170)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15)
50	13 <i>H</i> -Dibenzo[ <i>a,i</i> ]fluorene	239-60-1	1957 Lyons and Johnston (29)
			1958 British Empire Cancer Campaign (112)
			1959 Johnstone and Plimmer (63)
			1960 Hubert-Habart (153)
			1962/63 Pyriki (32)
			1963 Philip Morris (16)
51	Dihanzala alpantithacana	216 00 2	1964 Elmenhorst and Reckzeh (18)
51	Dibenzo[a,c]naphthacene	216-00-2	1957 Lyons and Johnston (29)
			1958 British Empire Cancer Campaign (112)
			1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
52	Dibenzo[a,j]naphthacene	227-04-3	1957 Wynder and Wright (76)
			1958 Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
53	Dibenzo[ <i>de,qr</i> ]naphthacene	193-09-9	1960 Rodgman and Cook (3),
54	Dibenzopyrene	58615-36-4	1959 Bonnet and Neukomm (61)
55	Fluoranthene	206-44-0	1954 Cooper and Lindsey (69)
			1955 Cooper and Lindsey (71), Lindsey (72)
			1956 Campbell and Lindsey (74), Gilbert and Lindsey (59, 60), Rodgman (1), Wright and Wynder (82)
			1957 Campbell and Lindsey (75), Lyons and Johnston (29), Wynder and Wright (76)
			1958 Ahlmann (83), , Pietzsch (84), Scassellati-Sforzolini and Salucci (85), Van Duuren (114, 120, 140), Van Duuren and Nelson (115), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Lindsey (78, 86), Lindsey et
			<i>al.</i> (144), Pietzsch (64), Wynder and Hoffmann (103, 110) Zapior <i>et al.</i> (65)
			1960 Candeli et al. (87), Hoffmann and Wynder (104, 105),
			Rodgman and Cook (3) 1960/61 Wynder (90)
			1961 Scassellati-Sforzolini and Mariani (92, 93), Scassellati-
			Sforzolini and Saldi (94) 1962 Lyons (66), Pietzsch (96)
			1963 Cook <i>et al.</i> (67), Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), Kröller (101), USPHS (15)
56	Fluoranthene, alkyl-		1958 Van Duuren (140), Van Duuren and Nelson (115)
	······································		1959 Wynder and Hoffmann (103)
			1960 Hoffmann and Wynder (105), Rodgman and Cook (3)
			1960/61 Wynder (90)
			1964 Elmenhorst and Reckzeh (18)
57	Fluoranthene, dimethyl-	60826-74-6	1958 Van Duuren and Nelson (115)
	······································		1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
58	Fluoranthene, 8,9-dimethyl-	25889-63-8	1963 Philip Morris (16)
			1959 Bentley and Berry (113)
59	Fluoranthene, 8-methyl-	20485-57-8	1958 Van Duuren (114, 120, 140), Van Duuren and Nelson
	·····		(115)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
60	9H-Fluorene	86-73-7	1956 Bonnet and Neukomm (58), Gilbert and Lindsey (59, 60)
			1957 Bonnet and Neukomm (28), Neukomm and Bonnet (108)
			1958 Ahlmann (83), Pietzsch (84)
			1959 Johnstone and Plimmer (63), Pietzsch (64)
			1960 Rodgman and Cook (3)
			1962 Lyons (66), Pietzsch (96)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18), USPHS (15)
61	9H-Fluorene, 1-methyl-	1730-37-6	1963 Philip Morris (16)
	· · · · ·		1964 Elmenhorst and Reckzeh (18)
62	9H-Fluorene, 9-methyl-	2523-37-7	1956 Bonnet and Neukomm (58)
	-		1957 Neukomm and Bonnet (108)
			1958 Van Duuren (120)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
63	1 <i>H</i> -Indene	95-13-6	1963 Cook <i>et al.</i> (67), Philip Morris (16)
64	Indeno[1,2,3-cd]fluoranthene	193-43-1	1960 Hoffmann and Wynder (105)
			1964 Elmenhorst and Reckzeh (18)
65	11 <i>H</i> -Indeno[2,1- <i>a</i> ]phenanthrene <sup>h</sup>	220-97-3	1958 Ahlmann (83)
			1959 Johnstone and Plimmer (63)
66	Indeno[1,2,3- <i>cd</i> ]pyrene	193-39-5	1960 Hoffmann and Wynder (105)
			1964 Elmenhorst and Reckzeh (18)
			1964 Elmenhorst and Reckzeh (18)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
67	Naphthacene	92-24-0	1962 Pietzsch (96)
			1963 Cook <i>et al.</i> (67)
			1964 Elmenhorst and Reckzeh (18)
68	Naphthalene	91-20-3	1954 Commins <i>et al</i> . (80)
	-		1955 Seelkopf (81)
			1956 Bonnet and Neukomm (58), Gilbert and Lindsey (59, 60)
			Rodgman (1)
			1957 Bonnet and Neukomm (28), Lyons and Johnston (29)
			1958 Pietzsch (84), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Pietzsch (64)
			1960 Rodgman and Cook (3)
			1962 Lyons (66)
			1963 Johnstone and Quan (171), Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
69	Naphthalene, alkyl-	4070 40 4	1963 Johnstone and Quan (171)
70	Naphthalene, 1,2-dihydro-4-methyl-	4373-13-1	1960 Rodgman and Cook (3)
71	Naphthalene, 1,6-dimethyl-	575-43-9	1963 Johnstone and Quan (171)
-		500 44 5	1964 Elmenhorst and Reckzeh (18)
72	Naphthalene, 1,8-dimethyl-	569-41-5	1956 Bonnet and Neukomm (58)
			1959 Bentley and Berry (113), Johnstone and Plimmer (63)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
73	Naphthalene, 2,6-dimethyl-	581-42-0	1963 Johnstone and Quan (171)
			1964 Elmenhorst and Reckzeh (18)
74	Naphthalene, 2,7-dimethyl-	582-16-1	1963 Johnstone and Quan (171)
			1964 Elmenhorst and Reckzeh (18)
75	Naphthalene, 1-methyl-	90-12-0	1952 Soffer and Stewart (39)
			1960 Rodgman and Cook (3)
			1961 Guérin (157)
			1963 Cook et al. (67), Johnstone and Quan (171)
			1964 Elmenhorst and Reckzeh (18)
76	Naphthalene, 2-methyl-	91-57-6	1955 Cooper and Lindsey (71)
			1956 Gilbert and Lindsey (59)
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1963 Cook <i>et al.</i> (67), Johnstone and Quan (171), Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
77	Naphthalene, 2-phenyl-	612-94-2	1960 Rodgman and Cook (3)
78	Naphthalene, 1,3,6-trimethyl-	3031-08-1	1963 Cook et al. (67), Johnstone and Quan (171)
			1964 Elmenhorst and Reckzeh (18)
79	1H-Naphtho[3,2,1,8-defg]chrysene	190-99-8	1957 Wynder ()
• -			1959 Johnstone and Plimmer (63)
80	Naphtho[2,1,8-qra]naphthacene	196-42-9	1957 Lyons and Johnston (29), Wynder and Wright (76)
			1958 Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Tennessee Eastman (116)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
81	Naphtho[1,2-b]triphenylene	215-26-9	1957 Lyons and Johnston (29)
			1958 British Empire Cancer Campaign (112)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
82	Pentaphene	222-93-5	1957 Wynder and Wright (76)
	- ·····	00 0	1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
83	Perylene	198-55-0	1956 Gilbert and Lindsey (59), Wright and Wynder (82) 1957 Lyons and Johnston (29)
			1958 British Empire Cancer Campaign (112), Pietzsch (84), Va Duuren and Nelson (115), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63), Pietzsch (64), Wynder and Hoffmann (103, 110)
			1960 Hoffmann and Wynder (104, 105), Rodgman and Cook (3 1960/61 Wynder (90)
			1962 Lyons (66), Pietzsch (96)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
84	Phenanthrene	85-01-8	1954 Cooper and Lindsey (69)
			1955 Cooper and Lindsey (71), Lindsey (72)
			1956 Bonnet and Neukomm (58), Gilbert and Lindsey (59, 60), Rodgman (1), Wright and Wynder (82)
			1957 Bonnet and Neukomm (28), Campbell and Lindsey (75), Neukomm and Bonnet (108), Rand <i>et al.</i> (109)
			1958 Ahlmann (83), Pietzsch (84), Scassellati-Sforzolini and Salucci (85), Wynder <i>et al.</i> (77)
		85-01-8	1959 Johnstone and Plimmer (63), Pietzsch (64)
			1960 Rodgman and Cook (3)
			1960/61 Wynder (90)
			1961 Scassellati-Sforzolini and Mariani (92, 93), Scassellati- Sforzolini and Saldi (94)
			1962 Lyons (66), Parmele and Jensen (95), Pietzsch (96)
			1963 Cook <i>et al</i> . (67), Philip Morris (16)
85	Phenanthrene, dimethyl-	29062-98-4	1964 Elmenhorst and Reckzeh (18), Kröller (101), USPHS (15 1956 Bonnet and Neukomm (58)
			1960 Rodgman and Cook (3)
86	Phenanthrene, 2,5-dimethyl-	3674-66-6	1958 Ahlmann (83)
			1959 Johnstone and Plimmer (63)
			1963 Philip Morris (16)
87	Phononthrono mothyl	31711-53-2	1964 Elmenhorst and Reckzeh (18) 1960 Rodgman and Cook (3)
88	Phenanthrene, methyl- Phenanthrene, 9-methyl-	883-20-5	1956 Bonnet and Neukomm (58)
			1957 Neukomm and Bonnet (108)
			1958 Ahlmann (83)
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
89	Pyrene	129-00-0	1953 Cooper and Lindsey (68)
			1954 Commins <i>et al.</i> (80), Cooper and Lindsey (69) 1955 Cooper and Lindsey (71), Lettré and Jahn (107), Lindsey
			(72), Lyons (73) 1956 Bonnet and Neukomm (58), Campbell and Lindsey (74), Gilbert and Lindsey (60), Rodgman (1), Wright and
			Wynder (82), Wynder and Wright (130)
			1957 Bonnet and Neukomm (28), Campbell and Lindsey (75), Lyons and Johnston (29), Neukomm and Bonnet (108),
			Rand <i>et al.</i> (109), Wynder and Wright (76) 1958 Ahlmann (83), Orris <i>et al.</i> (10), Pietzsch (84), Van Duurer (114, 120, 140), Van Duuren and Nakan (115), Wunder
			(114, 120, 140), Van Duuren and Nelson (115), Wynder et al. (77)
			1959 Bonnet and Neukomm (61, 62), Johnstone and Plimmer (63), Lindsey (78, 86), Lindsey <i>et al.</i> (144), Pietzsch (64) Wynder and Hoffmann (103, 110)
			1960 Candeli <i>et al.</i> (87), Cuzin (88), Hoffmann and Wynder (104, 105), Mouron <i>et al.</i> (89), Rodgman and Cook (3),
			Takayama and Oota (155) 1960/61 Wynder (90)

No.	Polycyclic Aromatic Hydrocarbon	CAS No.	References
89	Pyrene ( <i>cont</i> .)	129-00-0	1961 Mouron <i>et al.</i> (158), Sasmoco (91), Scassellati-Sforzolini and Mariani (92, 93), Scassellati-Sforzolini and Saldi (94)
			1962 Lyons (66), Parmele and Jensen (95), Pietzsch (96)
			1963 Cuzin <i>et al</i> . (97), Philip Morris (16)
			1964 USPHS (15)
90	Pyrene, alkyl-		1959 Wynder and Hoffmann (103, 110)
			1960 Hoffmann and Wynder (104, 105)
			1960/61 Wynder (90)
			1962 Lyons (66), Parmele and Jensen (95)
			1963 Philip Morris (16)
91	Pyrene, ethyl-		1955/56 Falk and Kotin (124)
00	Duran a matter t		1963 Philip Morris (16)
92	Pyrene, methyl-	27577-90-8	1958 Ahlmann (83), Van Duuren (120,140), Van Duuren and Nelson (115)
93	Pyrene, 1-methyl-	2381-21-7	1955 Cooper and Lindsey (71)
			1956 Bonnet and Neukomm (58), Gilbert and Lindsey (59, 60)
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
94	Pyrene, 2-methyl-	3442-78-2	1956 Bonnet and Neukomm (58), Rodgman (1)
			1958 Van Duuren (120), Wynder <i>et al.</i> (77)
			1959 Johnstone and Plimmer (63)
			1960 Rodgman and Cook (3)
			1963 Philip Morris (16)
05	Duran a Aurosthad		1964 Elmenhorst and Reckzeh (18)
95	Pyrene, 4-methyl-	3353-12-6	1956 Bonnet and Neukomm (58)
			1958 Van Duuren (114, 120, 140), Van Duuren and Nelson (115)
			1959 Johnstone and Plimmer (63)
			1960 Hoffmann and Wynder (105), Rodgman and Cook (3)
			1963 Philip Morris (16)
			1964 Elmenhorst and Reckzeh (18)
96	Rubicene	197-61-5	1961 Izawa (169)
97	Triphenylene	217-59-4	1961 Izawa (169)

<sup>a</sup> By January 1964, the date of the report of the Advisory Committee to the Surgeon General, of approximately 2100 PAHs bioassayed in laboratory animals, 480 (22.9%) were found to be tumorigenic to some degree.

<sup>d</sup> PAH was discussed by the Advisory Committee (15) as a carbon black component not a cigarette MSS component. <sup>e</sup> The PAH originally defined as cholanthrene (1,2-dihydrobenz[*j*]aceanthrylene) by RODGMAN and COOK (3) may have been one of several benzocyclopentanthracenes later reported by SNOOK *et al.* who identified over 500 PAHs in MSS (18). However, SNOOK *et al.* did not identify cholanthrene in cigarette MSS (19). In subsequent listings, the listing of benzocyclopentanthracenes replaced the original listing of 1,2-dihydrobenz[*j*]aceanthrylene (cholanthrene) as a tobacco smoke component

<sup>f</sup> The PAH known as benz[e]acephenanthrylene was formerly known as benzo[b]fluoranthene.

<sup>9</sup> The PAH originally identified in cigarette MSS by WYNDER and WRIGHT (31), LYONS and JOHNSTON (29), LYONS (30), BONNET and NEUKOMM (28), RODGMAN (2), RODGMAN and COOK (3), and PYRIKI (32) as dibenzo[*a*,/]pyrene

{dibenzo[*def,p*]chrysene} on the basis of published UV spectral data was subsequently shown by LAVIT-LAMY and BUU-HOï (33) to be dibenz[*a,e*]aceanthrylene. The authentic dibenzo[*a,I*]pyrene {dibenzo[*def,p*]chrysene} was subsequently identified in MS by SNOOK *et al.* (19) at the USDA.

<sup>h</sup> The PAH known as 11H-indeno[2,1-a]phenanthrene was formerly known as 11H-naphtho[2,1-a]fluorene.

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