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Original Article

# PHARMACOKINETICS OF ZINC IN BROILER CHICKENS AFTER SINGLE INTRAINGLUVIAL ADMINISTRATION WITH ZINC ASPARTATE

Sofiya L. Ivanova, Dimitrichka J. Dimitrova<sup>1</sup>, Metodi H. Petrichev, Liliana I. Parvanova, Georgi Sl. Kalistratov, Lubomir T. Vezenkov<sup>2</sup>

Non-infectious Diseases Unit, National Diagnostic and Research Veterinary Medical Institute, Sofia ¹Department of Pharmacology, Animal Physiology and Physiological Chemistry, Faculty of Veterinary Medicine, Trakia University, Stara Zagora ²Department of Organic Chemistry, University of Chemical Technology and Metallurgy, Sofia

## **Corresponding Author:**

Sofiya L. Ivanova
Non-infectious Diseases Unit
National Diagnostic and Research
Veterinary Medical Institute
15A, Pencho Slavejkov Blvd.
Sofia, 1606
e-mail: sofiya ivanova.com@abv.bg

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## **Summary**

The pharmacokinetics of zinc was investigated in broiler chickens after single crop intubation of 50 mg/kg 5% zinc aspartate suspension in 2% carboxymethyl cellulose solution. Blood serum zinc concentrations were assayed on a biochemical analyzer. The pharmacokinetics of zinc was evaluated using two approaches - compartmental method and non-compartmental analysis using pharmacokinetic software (TopFit, v. 2.0). After the intraingluvial application, zinc was rapidly absorbed  $(t_{1/2abs}=0.1040.02 \text{ h})$  by the alimentary system of birds attaining  $C_{max}$  of 63.603.94 mol/ml by hour 0.77 (compartmental method) and C<sub>max</sub>=69.274.35 mol/ml by hour 0.92 h (non-compartmental method). It is characterized with a long biological half-life (t<sub>1/2</sub>) of 13.821.63 h (compartmental analysis) and 15.961.73 h (non-compartmental analysis) and long mean residence times (MRT) 20.122.35 h and 23.002.50 h, respectively. The distribution in blood and extracellular fluid was good as seen from Vd<sub>(area)</sub> values 0.770.05 l/kg (compartmental analysis) and 0.650.05 l/kg (non-compartmental analysis).

**Key words:** pharmacokinetics, chickens, zinc aspartate

## Introduction

Zinc is an essential mineral for plants, animals and microorganisms [1]. It has an effect on growth. Zinc is a cofactor of numerous enzymes (more than 300) involved in the synthesis and degradation of proteins, carbohydrates, lipids, as well as in synthesis and catabolism of RNA and DNA [2-4].

The major part of zinc in the organism of animals and men is found in the brain, muscles, bones, kidneys and liver, with highest concentrations in the prostate and eyes [5, 6].

It is included in the diets of poultry due to its essential role in a number of metabolic processes. In modern poultry industry, both organic and inorganic zinc compounds are utilised [7-9].

There is scientific evidence about difference in pharmacokinetics of organic and inorganic zinc compounds in different animal species [5, 10-16]. There is no information about zinc pharmacokinetics in poultry.

The purpose of the present research was to determine the pharmacokinetics of zinc after

intraingluvial application of zinc aspartate in broiler chickens.

### **Materials and Methods**

In this pharmacokinetic study, 6 Cobb 500 broiler chickens (equal number of both genders), 45 days of age, weighing 1.5700.054 kg were included. The chickens were divided by gender in groups of 3 in metal cages at ambient temperature of 22-24C and air humidity 52%. They were fed compound food for growing birds and water was provided *ad libitum*.

The birds were treated with zinc aspartate, which represents an organic chelate complex of zinc synthesised in the Department of Organic Chemistry at the University of Chemical Technology and Metallurgy, Sofia. Due to the fact that the synthesised drug was insoluble in water, a 5% working suspension in 2% carboxymethyl cellulose was prepared *ex tempore*. Zinc aspartate was applied once, at a dose of 50 mg Zn/kg using an elastic silicone probe after 16-hour fasting.

Blood samples of 1 ml each were collected in Eppendorf tubes after venepuncture of the left wing v. brachialis. They were obtained on hour 0 (before treatment) and on post treatment hours 0.17, 0.33, 0.50, 1, 2, 4, 6, 8, 10, 12 and 24 h, left at room temperature for blood serum separation, centrifuged on a refrigerating centrifuge at  $1500 \times g$  for 15 min. Separated sera were stored deep frozen at -18 C for 24 h. The assay of serum zinc concentrations was performed on a biochemical analyzer (BS-200, Mindray Co.) using a colorimetric assay at wavelength =560 nm. Test principle of the 5-Br-PAPS commercial

kit consisted in the formation of coloured complex between zinc and [2-(5-Bromo-2-pyridiylazo)-5-[N-n-propyl-N-(3-sulfopropyl)amino]phenol,disodium]] whose colour intensity was proportional to the amount of zinc in the tested sample.

Using specialised pharmacokinetics software (TopFit, v.2.0.) [17] we determined zinc pharmacokinetics after intraingluvial application of zinc aspartate to broilers. Pharmacokinetic parameters of zinc behaviour after intraingluvial application were determined by the compartmental model [18] and by means of noncompartmental analysis [19]. With the former method, pharmacokinetic parameters were determined according to the Akaike information criterion (AIC) [20]. Pharmacokinetic parameters of zinc absorption, distribution and elimination were as follows: k<sub>el</sub> elimination rate constant;  $k_{abs.}$  absorption rate constant;  $T_{1/2}$ biological half-life; T<sub>1/2abs.</sub> absorption half-life; MRT mean residence time; Vd<sub>(area)</sub> volume of distribution; T<sub>lag</sub> hidden time; AUC<sub>0LOQ</sub> area under the serum concentration-time curve from time zero to the limit of quantitation; AUC<sub>0</sub> area under the serum concentration-time curve from time zero to infinity; C<sub>max</sub> maximum serum concentration;  $T_{max}$  time to reach  $C_{max}$ .

All pharmacokinetic values were presented as mean standard error of the mean (SEM).

#### Results

After single intraingluvial treatment with zinc aspartate, serum zinc concentrations were detected as early as the first sampling of blood in all birds (0.17 h) (Table 1).

**Table 1.** Serum zinc concentrations after oral treatment of broiler chickens with 5% zinc aspartate solution at a dose of 50 mg Zn/kg of body weight

Time interval (h)	Mean	SEM
0.17	20.783	1.699
0.33	43.233	2.404
0.50	54.817	3.717
_1	60.233	4.418
2	68.983	4.399
_4	60.117	3.491
_6	49.817	2.336
8	46.367	2.267
10	42.517	1.909
12	38.683	1.626
24	34.483	1.593

The follow-up of individual serum curves in broiler chickens showed that they fitted the onecompartmental open pharmacokinetic model.

The pharmacokinetic parameters after non-venous route of application determined by the compartmental method are presented in Table 2.

Pharmacokinetic parameters of zinc absorption from the gut of chickens after oral application of the zinc organic chelate complex are shown in Table 2 and include the time for its appearance in blood or hidden time  $(t_{lag})$  and absorption half-life  $(T_{1/2abs.})$ . Using both pharmacokinetic models (compartmental and non-compartmental analysis), maximum serum

concentrations ( $C_{max}$ ) of zinc were high and were attained soon after the intraingluvial application of zinc aspartate (Table 2).

After absorption, zinc was rapidly distributed in blood ensuring high values of the areas under the serum concentration-time curves (AUC $_0$ ; AUC $_{0LOQ}$ ).

Table 2 also presents pharmacokinetic parameters characterising the residence and elimination of zinc MRT and  $T_{1/2}$ . It could be seen that biological half-life and mean residence time of zinc after intraingluvial application of the chelate complex were long.

**Table 2.** Pharmacokinetic parameters of zinc after single oral treatment of broiler chickens with 5% zinc aspartate solution at a dose of 50 mg Zn/kg of body weight

Parameter	Units	Compartmental analysis		Non-compartmental analysis	
		Mean	SEM	Mean	SEM
T <sub>1/2abs</sub> .	h	0.104	0.015	-	-
T <sub>lag.</sub>	h	0.039	0.020	-	-
Τ <sub>1/2β</sub>	h	13.82	1.629	15.96	1.730
MRT	h	20.12	2.352	23.000	2.503
Vd (area)	l/kg	0.773	0.051	0.647	0.050
AUC $_{0\rightarrow \text{LOQ}}$	μmol x h/l	-	-	847.062	59.262
AUC <sub>0→∞</sub>	μ <b>mol x h/l</b>	1280.00	98.691	1211.837	225.354
T <sub>max</sub>	h	0.767	0.073	0.917	0.083
C <sub>max</sub>	μmol /l	63.600	3.941	69.267	4.346
r <sup>2</sup>	-	-	-	0.974	0.003

## **Discussion**

After intraingluvial treatment with zinc aspartate, serum concentrations of zinc were detectable in all blood samples from hour 0.17 to hour 24, similar to results from a previous study of ours on broilers treated with the organic compound zinc methionate at the same dose and concentration [21].

After application of zinc aspartate, serum zinc curves of all birds fitted the one-compartmental open pharmacokinetic model comparable to data reported for broilers by other researchers [8, 21] but differed from data registered in rats [5], rabbits [22, 23], fish [24], and humans [25] which fitted the two-compartmental pharmacokinetic model.

Data from Table 2 referring to the absorption half-life  $(T_{1/2abs.})$  indicated that, in chickens,

organic zinc was rapidly absorbed and consequently, maximum serum concentrations were rapidly attained. A similar tendency was established in an earlier study of ours in broiler chickens treated intraingluvially with another organic zinc compound – zinc methionate, applied at the same dose [21]. Unlike chickens, dogs treated orally with the zinc compound at 1 mg/kg exhibited twice longer  $T_{max}$  ( $T_{max}$ =2 h) [26]. A similar trend for more time needed for maximum serum concentrations were reported in humans by Nève, J. et al. (1991)  $T_{max}$ =2.3 h and in rabbits ( $T_{max}$ =1.33 h) [5].

After its occurrence in blood, zinc was rapidly distributed in the extracellular fluid and tissues with apparent volume of distribution Vd<sub>(area)</sub> 0.773 and 0.647 l/kg as per the two pharmacokinetic models, respectively (Table 2).

The pharmacokinetic profile of zinc after

single intraingluvial application of zinc aspartate is characterised with long-term residence of zinc in blood serum of broiler chickens, as indicated by biological half-life ( $T_{1/2}$ ) and mean residence time (MRT) values as per the two used models (Table 2).  $T_{1/2}$  and MRT values of zinc aspartate in broiler chickens were comparable to those reported by Ivanova, S. et al. (2014)[21] after single application of zinc methionate in the crop ( $T_{1/2}$ =15.45 h; MRT=22.50 h). Unlike chickens, the biological half-life of zinc in fish was considerably longer ( $T_{1/2}$ =16 days) as was the mean residence time (MRT=21 days) [24].

### **Conclusion**

From the results obtained it can be concluded that zinc in the form of organic chelate complex is rapidly absorbed, which is a prerequisite for rapid maximal serum concentrations of the metal. Zinc is rapidly distributed in the extracellular fluid and tissue in large volume, providing temporally for a high area under the curve of serum concentration-time ( $AUC_0$ ).

The serum pharmacokinetic profile shows monoexponential presentation and is characterized by long biological half-life ( $T_{1/2}$ ) and mean residence time (MRT), which explain the delayed elimination of zinc from the body of the birds.

### References

- 1. Sugarman B. Zinc and Infection. Rev Infect Dis.1983;5(1):137-47.
- Ashmead HD, Zunino H. Factors which affect the intestinal absorption of minerals, In: Ashmead HD, editor. The Roles of Amino Acid Chelates in Animal Nutrition. USA: Park Ridge: Noyes Publications; 1993. p. 21-46.
- 3. Kaim W, Schwederski B, Klein A. Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life: An Introduction and Guide. 2nd ed. Chichester: John Wiley & Sons Ltd.; 2013.
- 4. Panel on Micronutrients, Subcommittees on Upper Reference Levels of Nutrients and of Interpretation and Uses of Dietary Reference Intakes, and the Standing Committee on the Scientific Evaluation of Dietary Reference Intakes. Dietary Reference Intakes for Vitamin A, Vitamin K, Arsenic, Boron, Chromium, Copper, Iodine, Iron, Manganese, Molybdenum, Nickel, Silicon, Vanadium, and Zinc. Washington, D.C.: Food and Nutrition Board Institute of Medicine,

- National Academic Press; 2001.
- Yasuno T, Okamoto H, Nagai M, Kimura S, Yamamoto T, Nagano K, et al. The disposition and intestinal absorption of zinc in rats. Eur J Pharm Sci. 2011;44:410-15.
- García-Contreras Y, De Loera Y, García-Artiga C, Palomo A, Guevara JA, Herrrera-Haro J, et al. Elevated dietary intake of Zn-methionate is associated with increased sperm DNA fragmentation in the boar. Reprod Toxicol. 2011;31(4):570-3.
- 7. Yu Y, Wang RL, Xi L, Liu XG. Effects of zinc source and phytate on zinc absorption by in ligated intestinal loops of broilers. Poult Sci. 2010;89(10):2157-65.
- 8. Star L, van der Klis JD, Rapp C, Ward TL. Bioavailability of organic and inorganic zinc sources in male broilers. Poult Sci. 2012;91(12):3115-20.
- 9. Schlegel P, Sauvant D, Jondreville C. Bioavailability of zinc sources and their interaction with phytates in broilers and piglets. Animal. 2013;7(1):47-59.
- Rajas LX, McDowell LR, Martin FG, Wilkonson NS, Johnson AB, Njeru CA. Relative biavailability of zinc methionine and two inorganic zinc sources fed to cattle. J Trace Elem Med Biol. 1996;10(4):205-9.
- 11. Walter A, Krämer K, Most E, Pallauf J. Zinc availability from zinc lipoate and zinc sulphate in growing rats. J Trace Elem Med Biol. 2002;16(3):169-74.
- Wright CL, Spears JW, Webb JrKE. Uptake of zinc from zinc sulphate and zinc proteinate by ovine ruminal epithelia. J Anim Sci. 2008;86(6):1357-63
- 13. Li C H, Shen CC, Cheng YW, Huang SH, Kao CC, Liao JW, Kang JJ. Organ biodistribution, clearance, and toxicity of orally administered zinc oxide nanoparticles in mice. Nanotoxicology. 2012;6(7):746-56.
- 14. Pal DT, Gowda NK, Prasad CS, Amarnanth R, Bharadwa U, Suresh Babu G, et al. Effect of copper- and zinc-methionine supplementation on bioavailability, mineral status and tissue concentrations of copper and zinc in ewes. J Trace Elem Med Biol. 2010;24(2):89-94.
- 15. Chen J-K, Shin M-H, Peir J-J, Liu CH, Chou F-I, Lai W-H, et al. The use of radioactive zinc oxide nanoparticles in determination of their tissue concentrations following intravenous administration in mice. Analyst. 2010;135(7):1742-46.
- 16. Shyn A, Chalk SJ, Smith K, Charnock NL, Bielmyer GK. Zinc distribution in the organs of adult *Fundulus heteroclitus* after waterborne zinc exposure in freshwater and saltwater. Arch Environ Contam Toxicol .2012;63(4):544-53.
- 17. Heinzel G, Woloszak R, Thomann P. Topfit v.2.0.

- Pharmacokinetic and Pharmacodynamic Data Analysis System for PC. Stuttgart, Jena, New York: Gustav Fisher; 1993.
- 18. Baggot DJ. The Physiological Basis of Veterinary Clinical Pharmacology. Oxford (UK): Blackwell Science Ltd; 2001.
- 19. Gibaldi M, Perrier D. Pharmacokinetics, Revised and Expanded. 2nd ed. Swarbrick J, editor. New York: Informa Healthcare Inc; 2007.
- 20. Yamaoka K, Nakagawa T, Uno T. Application of Akaike's Information Criterion (AIC) in the evaluation of linear pharmacokinetic equations. J Pharmacokinet Biopharm. 1978;6:166-75.
- 21. Ivanova S, Dimitrova D, Petrichev M, Parvanova L, Kalistratov G, Vezenkov L. Pharmacokinetics of some inorganic ad organic zinc compounds in broiler chickens. Agric Sci Technol. 2014;6(3):267-70.
- 22. Andermann G, Dietz M. The bioavailability and pharmacokinetics of three zinc salts: zinc pantothenate, zinc sulphate and zinc orotate. Eur J Drug Metab Pharmacokinet. 1982;7(3):233-9.

- 23. Guillard O, Courtois P, Murai P, Ducassou D, Reiss D. Comparative pharmacokinetics of [65Zn] zinc sulphate and [65Zn] zinc pantothenate injected intravenously in rabbits. J Pharm Sci. 1984;73(11):1642-8.
- 24. Barron MG, Schltz IR, Newman MC. Pharmacokinetics of Intravascularly Administered <sup>65</sup>Zinc in Cannel Catfish (*Ictalurus punctatus*). Ecotoxicol Environ Saf. 2000;45:304-9.
- 25. Nève J, Hanq M, Peretz A, Abi Khalil F, Pelen E, Famaey JP, et al. Pharmacokinetic study of orally administered zinc in humans: Evidence for an enteral recirculation. Eur J Drug Metab Pharmacokinet. 1991;16(4):315-23.
- 26. Van der Broek AH. A standardized oral zinc tolerance test for assessment of zinc absorption in dogs. Vet Res Commun. 1993;17(1):3-11.