

A STUDY OF GROUNDWATER CHARACTERISTICS BY USING OF GROUNDWATER QUALITY INDEX AND POLLUTION INDEX IN ZUBAIR DISTRICT, BASRA PROVINCE, IRAQ

Ammar S. Dawood ^{a*}, Mushtak T. Jabbar ^{b*}, Mudhar H. Gatea ^a, Hayfaa J. Al-Tameemi ^c

^a Civil Engineering Department, College of Engineering, University of Basrah, Basra, Iraq

^b Geology /Earth Sciences/ HCC/ Seattle, Washington, 98198, USA

^c Department of Soil Science and Water Resources, University of Basrah, Basra, Iraq

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ABSTRACT:

The present work evaluated the groundwater quality index (GWQI) depending on some physicochemical analyses of thirteen groundwater samples in the Zubair district in Basra Province, Iraq. The collected groundwater samples were subjected to an extensive physicochemical analysis to evaluate the characteristics of water for drinking purpose according to Iraqi standard. For calculating the (GWQI), twelve water quality parameters were considered; turbidity, pH, chloride, total dissolved solids, total hardness (TH), electrical conductivity (EC), sodium, sulphate, phosphate, calcium, nitrate, and magnesium. The analysis of the results reveals that all the samples surpassed the portability of drinking water limits. High values of the (GWQI) in the obtained groundwater samples could possibly be caused by the higher values of electrical conductivity, total hardness, chloride, total dissolved solids and sodium within the groundwater. The calculated GWQI values ranged from 73.36 to 595.92. The pollution index (PI) was calculated for the study area with values ranged from 2.97 to 8.26. Correlation coefficients amongst the chosen water parameters exhibited some strong relationships. Finally, the analysis shows that the groundwater in this particular area needs to be treated before its consumption, and in addition, it usually needs to avoid the hazard of contamination. Principal component analysis (PCA) and cluster analysis (CA) indicate that the acquiring-data from groundwater samples are explained 90.5 % of the variance in the data with a four-component system that explains a large portion of the total variance of collected data.

1. INTRODUCTION

In all over the world, groundwater is used for many purposes including irrigation, domestic, moreover industrial uses. The pollution has acquired increasing in the last few decades due to the continuous-increment in population. There is an increase that is significant to the demand of the fresh water as a result of the rapid development in population and the expanded rate of progress in industrialization. Decreasing in water quality happens to be a worldwide issue of focus since human populations increase, agricultural and industrial activities enlarge, additionally climate change that jeopardizes to influence major changes to the hydrological-cycle (Federation and APHA, 2002). Relating to the WHO organization (WHO, 2004), nearly 80% of all the people's diseases are caused by water. Hence the quality of water must be expressed in the most common form to analyze the water characteristics.

Whenever groundwater is contaminated or degraded, its quality fails to recover by preventing the contaminants through the source. The guidelines and standards for drinking water quality

are planned to admit thoroughly clean and protected water distribution for human consumption, afterward protecting people's wellness. For that reason, it is a necessity to continuously monitor the groundwater quality and to protect it. The general target of any assessment for groundwater quality is often to get an all-inclusive description of the spatial distribution of the quality of groundwater and evaluate the changes in time that occur in the groundwater quality, either in a natural way or under the man's need (Tiwari and Nayak, 2002).

Water quality index (WQI) is the most considerable tool that is effective in conveying information about water quality towards the concerned citizens along with policy-makers. It really is a strategy which is efficient in determining the water characteristics (Singh 1992, Naik et al. 2001, Mishra et al. 2001). Water quality index, therefore, becomes a crucial parameter used in the management and assessment of groundwater. It can help in classifying groundwater: whether or not it is fit for drinking. WQI is computed from the standpoint

* Corresponding authors: Ammar S. Dawood, e-mail: ammars.dawood@yahoo.com;
Mushtak T. Jabbar, e-mail: mushtak1967@yahoo.com

of that groundwater suitability for man's consumption. WQI is distinctive as a score that reflects the composite-effect of a variety of water quality parameters. The calculation of the WQI in the groundwater study is started at the beginning with Horton (1965) and Landwehr (1974). Wu et al. (2011) reported that the picking of water quality parameters necessity evaluating of the primary anthropogenic-activity in the monitoring location. The primary anthropogenic activity may be domestic, agriculture, mining, etc. It is possible to determine the groundwater quality index (GWQI) by analyzing several important parameters and assign a proper weight for each one.

The aims of the current research are to assess the ground sources of water in the district of Zubair, as drinking water, by identifying the groundwater quality index and examine its suitability for man's consumption. The current work determines the levels of groundwater quality parameters depending on thirteen groundwater samples. Additionally, this research compares the determined levels with the Iraqi standard for drinking. Furthermore, the variability of parameters of the groundwater quality is explored in this paper by using multivariate statistical methods.

2. MATERIALS AND METHODS

2.1 Study area

The location of the study area is located in the Zubair district in the southern part of Basra Province, Iraq (Figure 1), between longitude 30.1672° N, 46.9805° E. In the year of 2008, Zubair district had a population of approximately 240,000 people (Wikipedia, 2017).

2.2 Data collection and sample analyses

Groundwater samples examined in this study were obtained from thirteen wells during 2014 by Department of Environmental Protection and Improvement in the Southern Region. The protocol of sampling process is done according to the internationally acceptable standards (Federation and APHA, 2002). Firstly, the physical parameters of water quality had been measured in situ for each sampling location. Secondly, the chemical parameters of water quality had been measured and assessed in the laboratory of the Department of Environmental Protection and Improvement in the Southern Region.

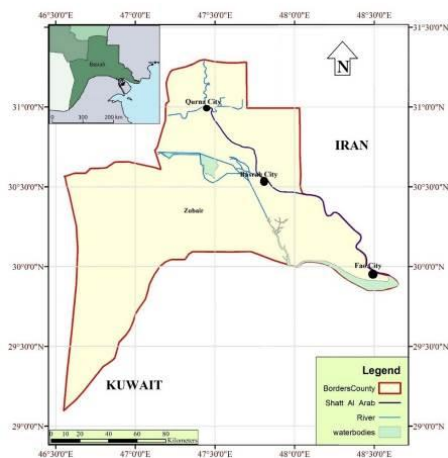


Figure 1. Basra Province map with Zubair district

The analysis of elements and parameters in the laboratory followed the standard methods. Each water-sample from the collected samples was assessed for eleven parameters such as turbidity, pH, chloride, sulphate, sodium, electrical conductivity, total hardness, total dissolved solids, magnesium, calcium, nitrate, and phosphate using standard-procedures of water test advised by the Federation and APHA (2002). The map for distribution of groundwater samples in the area of the study is displayed in Figure 2.

2.3 Data analysis

Descriptive-statistics of the parameters used for groundwater quality were carried out by implementing MS-Excel software version 2013 and making use of SPSS software version 20. The elements of descriptive-statistics for groundwater samples that result included mean, minimum, maximum, range, and standard deviation. Descriptive-statistics for the studied parameters for water quality of various sampling locations are introduced in Table 1.

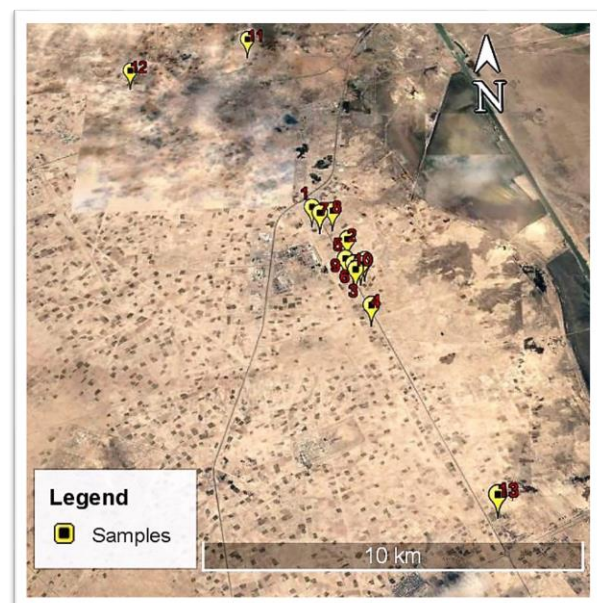


Figure 2. Location of groundwater samples in the area of the study in Zubair district

Table 1. Descriptive statistics of groundwater samples in the study area

Parameter	Mean	Min.	Max.	Range	Std.Dev
Turbidity	2.169	0.040	8.66	8.62	2.607
pH	7.292	7.000	7.60	0.6	0.189
Electrical Conductivity	8860.0	6010.0	15750.0	9740	2452.5
Total Dissolved Solids	6661.5	4420.0	13000.0	8580	2156.8
Chloride	1980.9	1140.0	4750.0	3610	948.8
Sulphate	1423.1	900.0	2000.0	1100	292.7
Sodium	1796.6	1100.0	4500.0	3400	905.9
Total Hardness	2571.1	1680.0	5600.0	3920	1099.4
Magnesium	245.8	132.0	579.00	447	140.6
Calcium	617.2	432.0	1280.00	848	216.9
Nitrate	35.7	0.156	53.79	53.6	18.263
Phosphate	0.65	0.310	2.68	2.37	0.640

2.4. Physicochemical parameters

The turbidity of the water is the haziness associated with the water as a result of the suspended individual particles. There are different sizes of suspended solid matter in the groundwater. The particles can effortlessly settle-down in water, meanwhile, the suspended solids remain in suspension and it appears as turbid water because it is not enough weight to get settled up. The measure of the activity for hydrogen ion in water solution is referred to as pH. In pure water and at a temperature equal to 25° C, the pH value close to 7. When the pH of solutions is less than 7, these types of solutions are generally referring to acidic solutions, but when the pH value of the solution is much larger than 7, it is classified as alkaline or basic solutions. WHO guidelines for drinking water quality state that the direct exposure, severe pH (on low or high pH values) effects of irritability to the eyes, mucous membrane, and skin of human beings.

Electrical conductivity is commonly applied as an index of the salt level and ion-contents contained in water. The features of pure-water are lower-conductivity and higher-resistivity. The number of inorganic substances that are obtained as the dissolved form is regarded as the total dissolved solids. The sizes of these inorganic substances (dissolved substances) tend to be even significantly lower than two micrometers and should not be identified in sieve tests.

In nature, chlorine is not present in free form and it most generally may appear as sodium chloride. The compounds of the chlorine are very soluble in water and therefore the groundwater gains chlorides by dissolving them in it. The compounds of the sodium are frequently found in the soil and rocks as a significant percentage. These compounds are dissolving in the groundwater by flow through the soil and rocks. One of the key elements for saltwater intrusion is the rise of sodium concentration in the groundwater. Other ions in water, such as sulfate, magnesium, calcium, nitrate, and phosphate, are also a significant component of groundwater which usually needed for drinking nonetheless in small amounts.

The water-hardness results from the existence of magnesium and calcium compounds along with other several minerals. Water is a good-solvent and efficiently dissolves the minerals that can come in contact. In water, the dissolved magnesium and calcium are majority typical minerals that make hardness of water.

2.5 Calculation of GWQI

Groundwater quality index (GWQI) is calculated in accordance with the following equation.

$$GWQI = \frac{\sum_{i=1}^n w_i q_i}{\sum_{i=1}^n w_i} \quad (1)$$

where,

w_i is express the unit weight of the i^{th} water quality parameter.

q_i is express the quality rating of the i^{th} water quality parameter.

The unit weight (w_i) of the i^{th} water quality parameter is computed in the subsequent equation and it is inversely

correspondent with the advised standards for the related parameters.

$$w_i = \frac{K}{S_i} \quad (2)$$

where,

K is express a constant-proportionality and it can be derived as in equation 3.

S_i is express the standard-value of the i^{th} water quality parameter.

$$k = \frac{1}{\sum_{i=1}^n (w_i)} \quad (3)$$

The quality rating is calculated according to the following equation

$$q_{ni} = \frac{(V_{actual} - V_{ideal})}{(V_{standard} - V_{ideal})} \times 100 \quad (4)$$

where,

q_{ni} is the quality rating of the i^{th} parameter for the total (n) number of the water quality parameters.

V_{actual} is the measured-value of water quality parameter (find from the laboratory).

V_{ideal} is the standard-value of water quality parameter (find from standard tables).

The value of V_{ideal} for pH is 7 and for the other studied water quality parameters is zero. Table 2, shows the water quality category depending on the computed value of GWQI.

Table 2. Water quality classification based on the computed value of WQI (Rupal et al, 2012)

Water Class	GWQI value	The status of water quality
I	< 50	Excellent
II	50-100	Good water
III	100-200	Poor water
IV	200-300	Very poor water
V	> 300	Unsuitable water for drinking

2.6 Calculation of Pollution index (PI)

The pollution index refers to the Nemerow pollution index, that is the method of calculation the pollution which prepared by Nemerow and Sumitomo in early 1970. Both of Ming et al. (2010) and Wu et al (2010) have studied this index. The simple form of the pollution index and it is specified as:

$$PI_i = \frac{C_i}{S_i} \quad (5)$$

where,

C_i is the measured-value of the i^{th} water quality parameter.

S_i is the standard-value of the i^{th} water quality parameter.

The average of pollution index (PI_{avg}) is defined as

$$PI_{avg} = \frac{1}{n} \sum_{i=1}^n PI_i \quad (6)$$

The perfect value of the PI_{avg} must be less than or equal to 1. There are four types of classification for the pollution-level which is dependent on water quality standard and it is represented in table 3.

Table 3. Level of pollution depending on water standard quality (Al-Othman, 2015)

Class of Water	PI_{avg} value	The status of pollution level
I	$0 < PI_{avg} < 1$	Good water
II	$1 < PI_{avg} < 5$	Slight polluted water
III	$5 < PI_{avg} < 10$	Medium pollution of water
IV	$PI_{avg} > 10$	Heavily polluted water

2.7 Correlation, PCA, and CA

The correlation is the research associated with the relationship between any two or more functionally variables. In water-quality studies, the analysis of correlations can be used to determine the forte and statistical significance of the relationship between any two or some other random water quality variables. The strength of the relationship between two random variables can be identified throughout the computation of correlation coefficient (R). When the value of r is close to zero, this means the poorer the correlation. The value of (R) ranges from -1 to 1. But when the value of (R) is close to -1, it implies that there is a strong negative correlation between the two random variables. But when the value of (R) is close to 1, it reveals that there is a strong positive correlation the two random variables, both variables decrease and increase with each other. The principal component analysis (PCA), cluster analysis (CA), and the correlation coefficient analysis (CCA), were used to determine the potential sources of the groundwater parameters.

3. RESULTS AND DISCUSSION

3.1 Results of physicochemical parameters

The Iraqi standard guideline for using water for drinking purposes and its related unit weights assigned are demonstrated in Table 3. From comparing the physiochemical results of the groundwater samples and its corresponding maximum values as recommended by the Iraqi standards that are shown in Table 1. It is noticeable that 38.5 % of the groundwater samples didn't abide by the Iraqi standards for NO_3 ; moreover, almost all the samples were not confirmed to the recommended EC, TDS, SO_4 , Na, TH, Mg, and Ca limits. Furthermore, all of the samples met the requirement for pH. Furthermore, 40% of groundwater samples were not compliant with the recommended PO_4 . Mean levels of EC, TDS, Cl, SO_4 , Na, TH, Mg, Ca and PO_4 were over the allowable levels of Iraqi standard for drinking water; obviously demonstrating the anthropogenic effect.

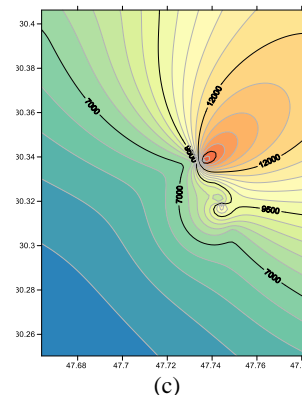
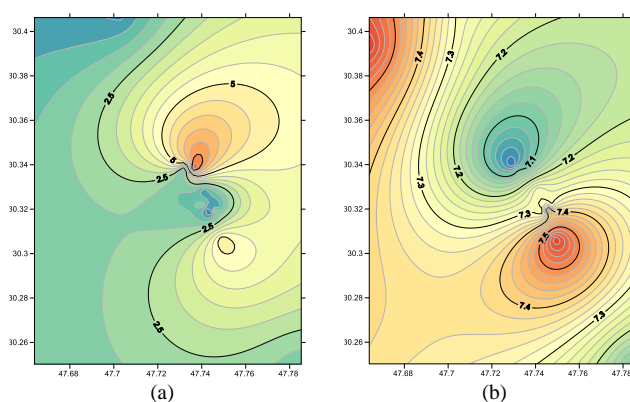


Figure 3. Spatial distribution map for (a) turbidity, (b) pH, and (c) electrical conductivity in the study area

The summary of analyzed-parameters result from this study-area is tabulated and presented in table 1 and Figs. 3-6, respectively. In study-area, it is noticed that the pH of samples lies within the allowable limit state by the Iraqi standard (6.5-8.5) with the average value of 7.292 (Table 1 and Figure 3.a). The value of turbidity in the study-area is ranged between 0.04 NTU to 8.66 NTU with the average value of 2.169 NTU (Table 1 and Figure 3.b). It is noticed from the results of the study-area that the electrical conductivity ranged between 6010 $\mu\text{S}/\text{cm}$ to 15750 $\mu\text{S}/\text{cm}$ with an average of 8860 $\mu\text{S}/\text{cm}$ (Table 1 and Figure 3.c). In the study-area, TDS values ranged between 1950-3150 mg/L with an average of 2269.7 mg/L (Table 1 and Figure 4.a). In groundwater, when the total dissolved solids have high value, it may be not harmful to people, but it has an effect on the persons who are suffering from heart diseases and kidney. The chloride concentration is high in groundwater, whenever rainfall is less and the temperature is high. The chloride content in the area of the study was revealed to be more than the allowable levels. The chloride ranges between 1140-4750 mg/L with an average value of 1980.92 mg/L (Table 1 and Figure 4.b). The permissible value of sulphate for drinking water is 400 mg/L. It is presented with the results of the area of the study that the concentration of sulphate ranged between 900-2000 mg/L with an average of 1423.077 mg/L (Table 1 and Figure 4.c). It is proven that all the groundwater samples of the study area exceed the permissible value of sulphate in drinking water recommended by the Iraqi standard.

Table 4. Iraqi standard guideline for drinking water and its corresponding unit weights (Hadeel, 2014)

Parameter	Standard value	Assigned unit weight
Tur.	5	0.2000
pH	8.5	0.1176
EC	1000	0.0010
TDS	1000	0.0010
Cl	350	0.0029
(c) SO_4	400	0.0025
Na	200	0.0050
TH	500	0.0020
Mg	100	0.0100
Ca	150	0.0067
NO_3	50	0.0200
PO_4	0.4	2.5000

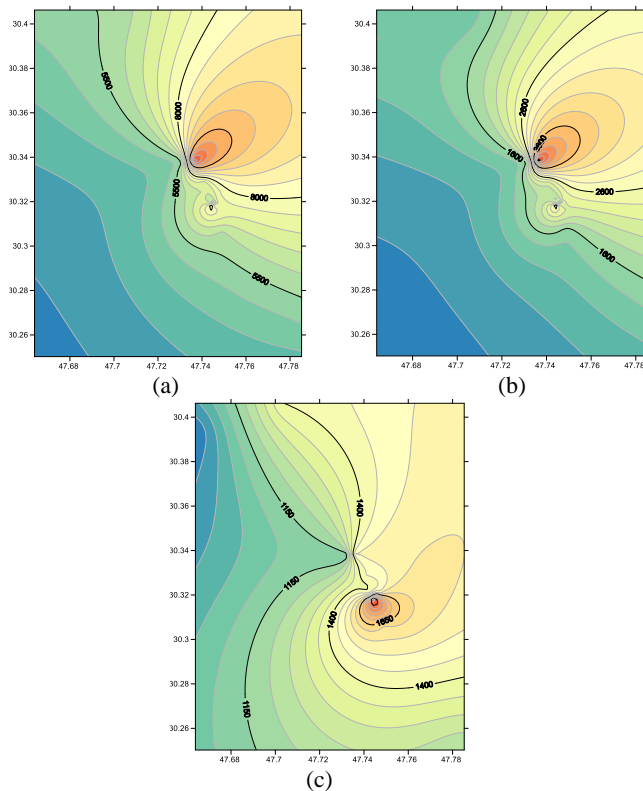


Figure 4. Spatial distribution map for (a) total dissolved solids, (b) chloride, and (c) sulphate in the study area

It is found from the analysis of the results that all the groundwater samples have a sodium-content more than the maximum limit of 200 mg/L depending on the Iraqi standard for drinking water. The range of sodium levels in the area of the study is 110-4500 mg/L (Table 1 and Figure 5.a). The hardness of water shows that it ranged between 1680 mg/L to 5600 mg/L (Table 1 and Figure 5.b). In the area of the study, the concentration of magnesium ranged between 132-579 mg/L with an average of 245.84 mg/L (Table 1 and Figure 5.c). The range of magnesium values in the area of the study is not permissible.

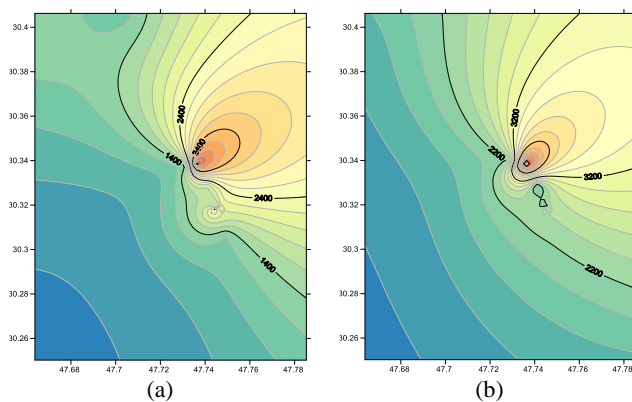
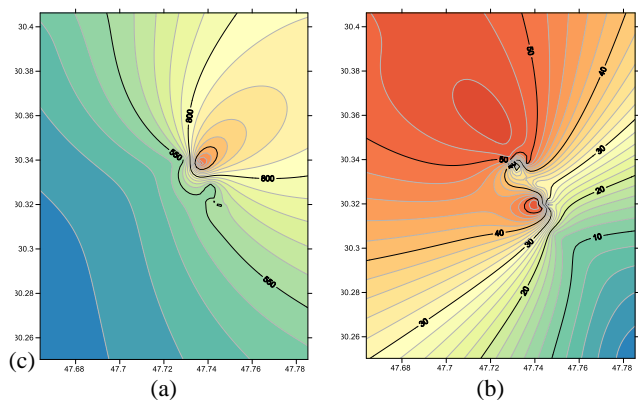


Figure 5. Spatial distribution map for (a) sodium, (b) total hardness, and (c) magnesium in the study area

The calcium content in groundwater samples of the area of the study ranged between 432-1280 mg/L with an average of 617.23 mg/L (Table 1 and Figure 6.a). The range of calcium values in the area of the study is not permissible because its values were exceeded the permissible value of calcium in drinking water recommended by the Iraqi standard. In study-area, the nitrate value varied from 0.156 mg/L to 53.79 mg/L (Table 1 and Figure 6.b). The value of nitrate content in the study area was found to be more than the value of 50 mg/L as per Iraqi standard in 6 locations. The high value of nitrate concentration is located in the rural part of the study area as a result of the over-use of fertilizer and the improper-operation as well as the maintenance of septic systems. The phosphate content ranges between 0.31-2.68 mg/L with an average value of 0.655 mg/L (Table 1 and Figure 6.c).

3.2 Results of GWQI calculations

The value of GWQI demonstrates its appropriateness for human-drinking. In this research, the calculated GWQI values ranged from 73.363 to 595.9186 and which means that the GWQI can be classified into three types as shown in Table 4 and Figure 7. The higher GWQI value means the most polluted in groundwater. The higher value of GWQI in these locations of the study-area has found from the higher-values of total hardness, nitrate, total dissolved solids, and also manganese in the groundwater-samples. Table 5 shows the GWQI of water-samples.



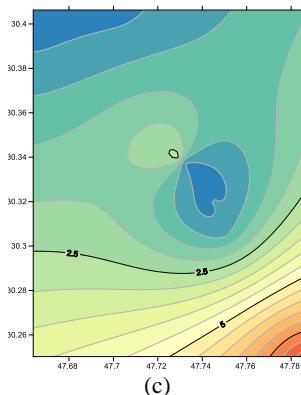


Figure 6. Spatial distribution map for (a) calcium, (b) nitrate, and (c) phosphate in the study area

Table 5. GWQI and PI values of the groundwater samples

Well	1	2	3	4	5	6	
GWQI	595.9	87.3	185.5	172.6	91.9	83.7	
PI	4.04	3.83	3.68	3.48	3.89	4.23	
Well	7	8	9	10	11	12	13
GWQI	112.5	146.5	75.4	73.4	75.9	83.5	195.4
PI	4.19	8.26	4.94	3.53	3.33	2.97	2.80

From Table 4 and Figure 7, the groundwater samples from all sampling locations had only one GWQI greater than 300 and can, for this reason, be regarded as unsuitable for drinking without previous treatment. The electrical conductivity, total dissolved salt and other ions of the groundwater samples are mainly the reason for the higher GWQI values. Approximately 54% of groundwater samples had GWQI values with the type of good water. Around 38% of groundwater samples had GWQI values with the poor water type.

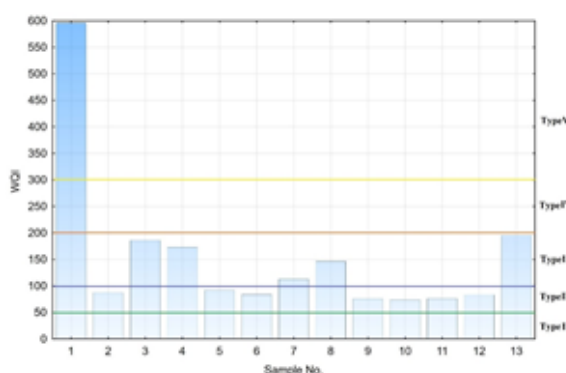


Figure 7. GWQI values for the groundwater samples in the area of the study

3.3 Results of pollution index

The average pollution index (PI_{avg}) values range from 2.97 to 8.26. From the results of PI_{avg} for the selected groundwater samples (Table 4), depending on PI_{avg} , 8% of the groundwater samples indicate medium polluted water and 92% of the groundwater samples have slightly polluted water (Figure 8). Figure 8 displays the variation of PI_{avg} at various locations of the study area.

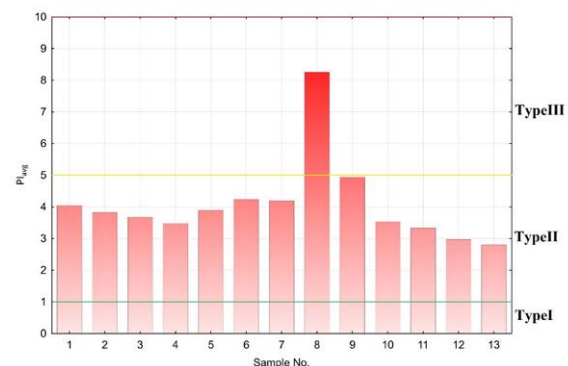


Figure 8. PI_{avg} values for the groundwater samples in the area of the study

3.4 Results of statistical analysis

Figure 9, shows the scatter plot that denotes the data is of a multi-component scheme and this really is revealed by Table 5 as four components cumulatively explained about 90.5 % of the variance in the collected data. From Table 5, exclusively components 1 to 4 has Eigenvalues which are greater than 1.

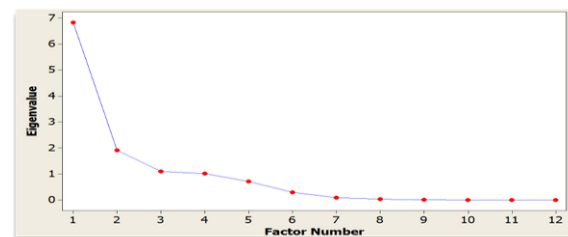


Figure 9. Scatter plot of Eigenvalues for the groundwater samples

From the Principal Component Matrix (Table 6), the representation for each PCs can be written as follow:

$$PC1 = 0.642 \text{ Tur} - 0.398 \text{ pH} + 0.931 \text{ EC} + 0.966 \text{ TDS} + 0.957 \text{ Cl} + 0.246 \text{ SO}_4 + 0.957 \text{ Na} + 0.911 \text{ TH} + 0.830 \text{ Mg} + 0.957 \text{ Ca} + 0.349 \text{ NO}_3 - 0.066 \text{ PO}_4 \dots\dots\dots (7)$$

$$PC2 = 0.385 \text{ Tur} - 0.642 \text{ pH} - 0.265 \text{ EC} - 0.222 \text{ TDS} - 0.162 \text{ Cl} - 0.684 \text{ SO}_4 - 0.145 \text{ Na} + 0.199 \text{ TH} + 0.249 \text{ Mg} + 0.135 \text{ Ca} - 0.132 \text{ NO}_3 + 0.765 \text{ PO}_4 \dots\dots\dots (8)$$

$$PC3 = -0.150 \text{ Tur} + 0.060 \text{ pH} - 0.182 \text{ EC} - 0.094 \text{ TDS} - 0.116 \text{ Cl} - 0.162 \text{ SO}_4 - 0.102 \text{ Na} + 0.315 \text{ TH} + 0.396 \text{ Mg} + 0.212 \text{ Ca} - 0.642 \text{ NO}_3 - 0.512 \text{ PO}_4 \dots\dots\dots (9)$$

$$PC4 = -0.467 \text{ Tur} - 0.176 \text{ pH} - 0.016 \text{ EC} - 0.039 \text{ TDS} - 0.017 \text{ Cl} - 0.503 \text{ SO}_4 + 0.005 \text{ Na} + 0.071 \text{ TH} + 0.123 \text{ Mg} + 0.014 \text{ Ca} + 0.614 \text{ NO}_3 - 0.336 \text{ PO}_4 \dots\dots\dots (10)$$

The standardization factors were substitution into expressions for the four PCs. The values for PCs 1,2,3 and 4 were discovered by determining each of PCs of water quality individually. Since PCs 3 and 4 are comparatively less than

PCs 1 and 2, accordingly it is more difficult to analyze it in the three-dimensional space, PCs 1 and 2 were selected for the distribution analyze of water quality in this research (see Figure 10).

The cluster analysis was utilized for the identification of the spatial-similarity amongst the sampling sites that depending on the levels of groundwater parameters. All of the thirteen sampling sites are grouped into four statistically significant clusters, as illustrated by the Dendrogram (see Figure 11). A Dendrogram is an important tool in assessing the significant variables and source of pollution. From Figure 10, only two sampling locations are spatially comparable, i.e. cluster 1 (1 and 7), six locations (2, 6, 5, 10, 11 and 9) form the second cluster while the four locations (3, 4, 12 and 13) form the third cluster. Finally, one location for the fourth cluster.

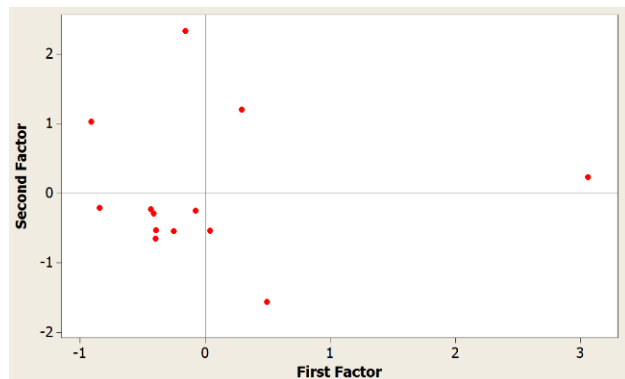


Figure 10. Values of PC1 and PC2 for water samples

The coefficients of Pearson correlation between the selected groundwater properties revealed high and strong relationships (see Table 7). It was noticed that there is a high correlation ($R > 0.9$) that were observed between the following parameters: (EC with each of TDS, Cl, and Na), (TDS with each of Cl and Na), (Ca with each of SO_4 and Mg), and (Cl and Na). Table 7 also shows a strong relationship ($R > 0.8$) between the following parameters: (EC and Ca), (TDS and Ca), (Cl and Ca), and (Ca and Na).

Table 6. Principal Component Matrix (PCM) of data (groundwater samples)

Variable	PCs1	PCs2	PCs3	PCs4
Tur.	0.642	0.385	-0.150	-0.467
pH	-0.398	-0.642	0.060	-0.176
EC	0.931	-0.265	-0.182	-0.016
TDS	0.966	-0.222	-0.094	-0.039
Cl	0.957	-0.162	-0.116	-0.017
SO_4	0.246	-0.684	-0.162	-0.503
Na	0.957	-0.145	-0.102	0.005
TH	0.911	0.199	0.315	0.071
Mg	0.830	0.249	0.396	0.123
Ca	0.957	0.135	0.212	0.014
NO_3	0.349	-0.132	-0.643	0.615
PO_4	-0.066	0.765	-0.512	-0.336
Variance	6.8265	1.9175	1.0945	1.0159
% Var	0.569	0.160	0.091	0.085

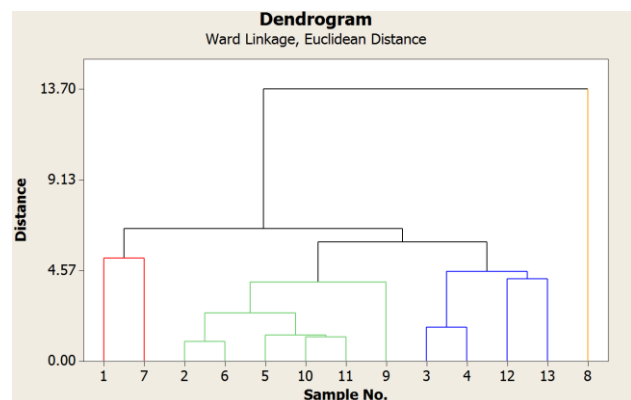


Figure 11. Dendrogram using Ward Linkage (between Groups)

Table 7. Correlation of groundwater quality parameters in the samples of the area of the study

Parameter	Turbidity	pH	EC	TDS	Cl	SO_4	Na	TH	Mg	Ca	NO_3	PO_4
Turbidity	1											
pH	-.195	1										
EC	.500	-.248	1									
TDS	.570	-.249	.977	1								
Cl	.554	-.309	.957	.985	1							
SO_4	.008	.222	.454	.405	.342	1						
Na	.559	-.312	.952	.982	.996	.299	1					
TH	.592	-.444	.728	.791	.775	.027	.780	1				
Mg	.505	-.441	.618	.686	.675	-.040	.676	.980	1			
Ca	.658	-.428	.815	.869	.849	.096	.858	.976	.914	1		
NO_3	.055	-.089	.445	.388	.377	-.034	.376	.170	.128	.210	1	
PO_4	.442	-.437	-.163	-.182	-.127	-.251	-.134	-.076	-.071	-.080	.004	1

4. CONCLUSIONS

The evaluation study of the groundwater quality index (GWQI) was carried out in Zubair district depending on the physicochemical analyses of thirteen groundwater samples in the study area. The quality of groundwater samples is obtained from the area under investigation that examined and analyzed of twelve parameters, specifically turbidity, pH, electrical conductivity, chloride, sulphate, sodium, total dissolved solids, total hardness, magnesium, calcium, nitrate, and phosphate. Furthermore, the studied parameters were used to obtain the GWQI and the average of PI values. Most groundwater samples exceeded the upper limit of the Iraqi standard for drinking water. GWQI values range from 73.36 to 595.92 and the PIavg values range from 2.97 to 8.26. The main objective of this research was to formulate a database map for the area of the study and to assess the quality of groundwater in Zubair district. The Pearson correlation coefficients between the selected groundwater properties showed different relationships. Principal component analysis and cluster analysis that were used in this study recommend the data is of a four-component system type which is explained approximately 90.5% of the total variance in the data used. The study indicates that the groundwater of the study area needs prior treatment before human-consumption such as drinking.

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