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THE EFFECT OF CADMIUM ON OXIDATIVE STRESS IN *Beta vulgaris*

WPLYW KADMU NA STRES OKSYDACYJNY U *Beta vulgaris*

Abstract: As a heavy metal, cadmium has strongly toxic effects on plants and can induce oxidative stress. It is absorbed by the roots and transported to the stems and leaves. The aim of the study was to evaluate the effect of various concentrations of cadmium on the metabolic activity of *Beta vulgaris* and assess the dependence of these processes on the content of metal in the plants. To demonstrate the effect of cadmium on metabolism, protein and photosynthetic pigment content, lipid peroxidation, and the activity of enzymes specific for oxidative stress in roots and shoots were measured. Seeds of *B. vulgaris* were treated with different concentrations of Cd supplied via a CdCl₂ solution: 0 (control), 200, 300 and 400 mg/dm³. Results of the present study revealed increased GPOX activity as cadmium concentration rose, while SOD activity was stimulated by a low Cd concentration (200 mg/dm³) and reduced by high levels of Cd. Based on the present findings, it can be concluded that GPOX in *B. vulgaris* played a more important role in ROS scavenging than SOD did and was able to reduce the level of lipid peroxidation in plants. Cadmium, in the concentration range used, did not show any significant effect on protein or photosynthetic pigment content.

Keywords: *Beta vulgaris*, cadmium, oxidative stress

Introduction

Heavy metals are elements naturally occurring in the Earth's crust. Many have strongly toxic effects [1, 2]. One of them is cadmium [3]. In the natural environment, cadmium is often associated with lead and zinc ores [4]. In noncontaminated soil, Cd concentration varies from 0.01 to 5.0 mg/kg [5]. However, it is estimated that about 30,000 Mg of cadmium are released into the environment every year. This pollution originates mainly from cement production, the metallurgical industry, fossil fuel combustion and the use of artificial fertilizers [6, 7].

For higher plants, cadmium has no biological function. It is absorbed by the roots and transported to the stems and leaves via the symplast and apoplast pathways [8, 9]. Plants activate various mechanisms in response to the presence of cadmium, including chelation, active transport into the vacuole, immobilization, and compartmentalization of metal ions. Stress protein expression and increased ethylene production have also been observed [5].

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The most general symptoms of Cd stress are declines in chlorophyll content and inhibited photosynthesis, which manifest as less growth and reduced leaf area [10, 11]. Furthermore, Cd stress increases reactive oxygen species (ROS) levels, which can lead to damage to vital molecules, such as proteins, lipids, DNA and RNA [12]. Those impairments can generate a metabolic malfunction in cells and, as a result, may cause cell death [13].

ROS are reactive atoms or molecules found in all aerobic organisms. Free radical ROS, including hydroxyl radicals, alkoxy radicals, perhydroxyl radicals and superoxide radicals, as well as non-radical forms (singlet oxygen and hydrogen peroxide), are naturally produced in the mitochondria and chloroplasts of plants and animals.

ROS overproduction causes oxidative stress and activates various defense mechanisms. Organisms use antioxidant enzymes, including ascorbate peroxidase, superoxide dismutase and catalase, to reduce the amount of ROS [13, 14]. Non-enzymatic defense mechanisms include glutathione, non-protein amino acids, phenolic compounds and ascorbic acid [14, 15].

The aim of this study was an experimental study of *Beta vulgaris*, measuring metabolic stress under the influence of cadmium in various concentrations and assessing the dependence of these processes on the content of metal in the plants.

To demonstrate the effect of cadmium on metabolism, protein and photosynthetic pigment content, lipid peroxidation, and the activity of enzymes specific for oxidative stress were determined.

Materials and methods

Plant materials and treatments

Seeds of *B. vulgaris* (7 g) were sterilized in a 5 % (w/v) H_2O_2 solution for 10 min and then treated with different concentrations of Cd supplied via a CdCl_2 solution: 0 (control), 200, 300 and 400 mg/dm^3 . Each treatment was done in three replicates. After 12 hours of Cd treatment, the seeds were washed with sterile water and sown in plastic pots filled with perlite. The cultivation was carried out in a growth chamber in a photoperiodic system day/night 14/10 hours at a temperature of 20/16 °C, respectively, with 80 % relative humidity. Irrigation via a Knap solution was provided regularly in sufficient quantities for plant growth.

Enzyme extractions and assay

The roots and shoots samples (1 g) were ground separately in liquid N_2 using a mortar and pestle, after which the ground material was homogenized on ice in 2 cm^3 of a 50 mM sodium phosphate buffer (pH = 7.0). The homogenates were centrifuged at 10,000 rpm (revolutions per minute) for 20 min at 4 °C. The supernatants were used in the determination of guaiacol peroxidase (GPOX, EC 1.11.1.7) and superoxide dismutase (SOD, EC 1.15.1.1) activity and protein content.

GPOX activity was determined spectrophotometrically using guaiacol as the substrate and H_2O_2 as the hydrogen donor. Colored tetraguaiacol ($\epsilon = 26.6 \text{ mM}^{-1}\text{cm}^{-1}$) is formed in guaiacol oxidation process. The specific GPOX activity was determined according to the modified method by Zaharieva et al. [16]. The reaction mixture (5.0 cm^3) consisted of a 50 mM phosphate buffer (pH = 7.0), 38 mM of H_2O_2 and 4 mM of guaiacol. The reaction was started by the addition of 0.2 cm^3 of the supernatant (enzyme extract). The increase in

absorbance was measured at 470 nm for 1-minute intervals up to 5 minutes. The enzyme activity was expressed in μmol per minute and milligrams of protein.

SOD activity was determined by its ability to inhibit autooxidation of adrenaline and thereby prevent the formation of active forms of oxygen [17]. To prepare this, 0.055 cm^3 of a 0.18 % (5.46 mmol) pharmacy solution of adrenaline hydrochloride was added to 2 cm^3 of a 0.2 M sodium carbonate buffer (pH = 10.65), then mixed thoroughly and rapidly. Absorption intensity was determined every 30 seconds for 5 minutes at a wavelength of 340 nm (A_1), 0.01 cm^3 of the enzyme extract and 0.055 cm^3 of 0.18 % adrenaline hydrochloride were added to 2 cm^3 of buffer (pH = 10.65) and stirred, and the absorption intensity was measured as described above (A_2). The blank was a buffered enzyme extract, without adrenaline. Antioxidant activity (AA) of the studied enzyme extracts were expressed as the percent inhibition of autooxidation of adrenaline and calculated by the formula:

$$AA = \frac{(A_1 - A_2) \cdot 100 \%}{A_1}$$

One unit of SOD activity was defined as the amount of enzyme that caused the inhibition of autooxidation of adrenaline by 50 %. Enzyme activity was expressed in units per minute and milligrams of protein.

Determination of protein content

The amount of total protein of the roots and shoots samples was measured using the Bradford method with bovine serum albumin (BSA) as the standard [18]. Absorption intensity of the extract was determined at a wavelength of 595 nm and the results were reported in mg/g f.m. (fresh mass).

Determination of photosynthetic pigments content

The content of photosynthetic pigments was determined by acetone extraction. Briefly, 0.1 g of roots and shoots samples were homogenized separately using a chilled mortar and pestle with 5 cm^3 of 80 % (v/v) acetone. The homogenates were centrifuged at 10,000 rpm for 10 min at 4 °C. The absorbance of the supernatant was measured at 647, 663 and 470 nm. The contents of chlorophyll *a*, chlorophyll *b* and carotenoids were calculated according to Lichtenthaler [19] and are given in mg/g f.m.

Determination of lipid peroxidation

Lipid peroxidation was expressed by the malondialdehyde (MDA) content, determined by the thiobarbituric acid (TBA) reaction described by Heath and Packer [20]. Briefly, 0.3 g of roots and shoots samples were homogenized separately in 4 cm^3 of 0.25 % (w/v) thiobarbituric acid (TBA) in 10 % (w/v) trichloroacetic acid (TCA) using a mortar and pestle. The homogenates were heated at 95 °C for 30 min, quickly cooled in an ice bath and then centrifuged at 10,000 rpm for 10 min. The absorbance of the supernatant was recorded at 532 nm and 600 nm. The blank was 0.25 % TBA in 10 % TCA. The MDA concentration was calculated by subtracting the absorbance at 600 nm (nonspecific turbidity) using the extinction coefficient of $155 \text{ mM}^{-1} \cdot \text{cm}^{-1}$ and expressed in terms of nmol/g f.m. [21].

Determination of cadmium content

The roots and shoots samples (the dry mass of each was 0.400 ± 0.001 g) were mineralized in a mixture of nitric (V) acid and hydrochloric acid (HNO_3 65 % : HCl 37 % = = 1 : 3) using a Speedwave Four microwave oven (Berghof, Germany). The mineralization process temperature was 180 °C. Reagents (Merck, USA) were used to prepare solutions [22]. Cadmium was determined with the atomic absorption spectrometer iCE 3000 (Thermo Electron Corporation, USA). The Instrument Detection Limit (*IDL*) and the Instrument Quantification Limit (*IQL*) for cadmium were 0.0028 and 0.013 mg/dm³, respectively. The highest concentration of the calibration standard available from ANALYTIKA Ltd. (Czech Republic), 2.5 mg/dm³, was adopted as the upper limit of the linear relation between the concentration of the analyte and the instrument signal [23]. The quality control of measurements was conducted using test analyses of the BCR 414 *plankton* and BCR-482 *lichen* reference materials (Institute for Reference Materials and Measurements, Belgium). The obtained results are summarized in Table 1.

Table 1

Measured and certified values of Cd concentration in the BCR 414 *plankton* and the BCR 482 *lichen* reference material

BCR 414 <i>plankton</i>					BCR 482 <i>lichen</i>				
Certified value	±Uncertainty	AAS		<i>D</i> *	Certified value	±Uncertainty	AAS		<i>D</i> *
		Mean	±SD				Mean	±SD	
[mg/kg d.m.]					[mg/kg d.m.]				
[%]					[%]				
0.383	0.014	n.d.	n.d.	n.d.	0.56	0.02	0.53	0.03	-5.3

* Deviation - the difference between the measured value and the certified value, divided by the certified value
n.d. - not detected

Statistical analysis

All experiments were carried out three times, with results expressed as mean values ± standard deviation. One-way analysis of variances (ANOVA) and t-Student's test were used to determine statistical differences between the treatment samples and the control using Excel (Microsoft, USA). Differences were considered significant at the level of $p < 0.05$. In order to assess the interdependence of cadmium content in *B. vulgaris* roots and shoots and the tested oxidative stress markers, a correlation analysis was performed using Excel (Microsoft, USA).

Results

Protein content

The effect of different Cd concentrations on protein content in the roots and shoots of *B. vulgaris* is shown in Figure 1. In the concentration range used, cadmium did not show any significant effect on protein content in either part of the plant.

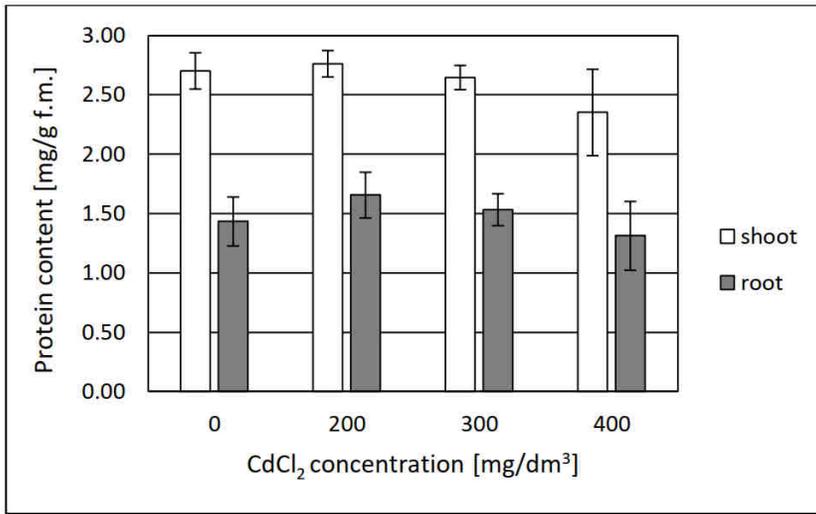


Fig. 1. The effect of different cadmium concentrations on protein content in *Beta vulgaris* plants (bars represent standard deviation); ANOVA, $p > 0.05$ (shoot), $p > 0.05$ (root)

Superoxide dismutase activity

The effect of different Cd concentrations on SOD activity in the roots and shoots of *B. vulgaris* is shown in Figure 2. The activities of this enzyme were higher in roots than shoots. For shoots, cadmium did not show any significant effect on SOD activity in the concentration range used. For roots, superoxide dismutase activity increased with a lower concentration of Cd, whereas a significant decrease was observed in response to higher concentrations of the metal (300 and 400 mg/dm³).

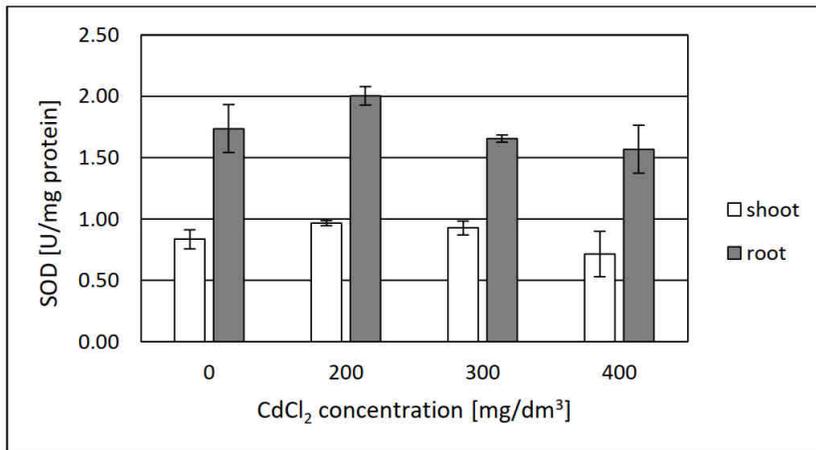


Fig. 2. The effect of different cadmium concentrations on the activity of superoxide dismutase (SOD) in *Beta vulgaris* plants (bars represent standard deviation); ANOVA $p > 0.05$ (shoot), $p < 0.05$ (root)

Guaiacol peroxidase activity

The effect of different Cd concentrations on GPOX activity in the roots and shoots of *B. vulgaris* is shown in Figure 3. The activity of the enzyme involved in the removal of H_2O_2 was higher in roots than in shoots. Furthermore, in both parts of the plant, the activity of GPOX significantly increased as cadmium concentration increased. In the case of roots, a higher concentration of Cd (400 mg/dm^3) slightly reduced GPOX activity.

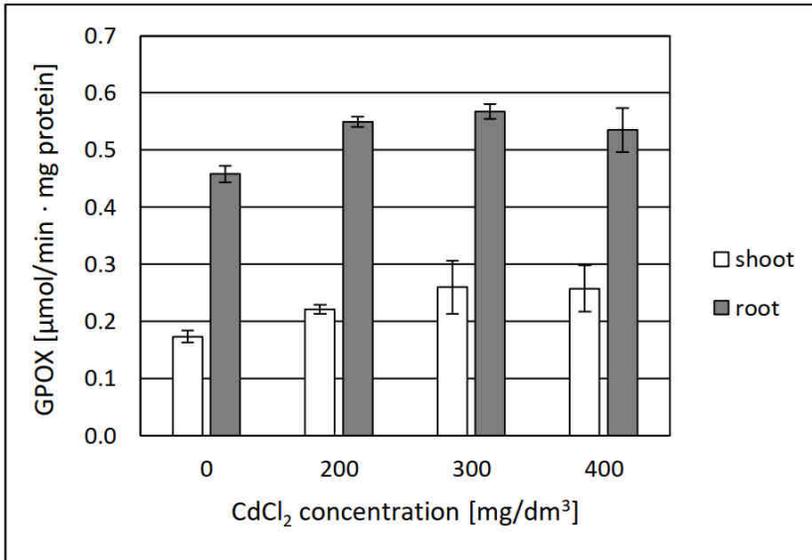


Fig. 3. The effect of different cadmium concentrations on the specific activity of guaiacol peroxidase (GPOX) in *Beta vulgaris* plants (bars represent standard deviation); ANOVA, $p < 0.05$ (shoot), $p < 0.05$ (root)

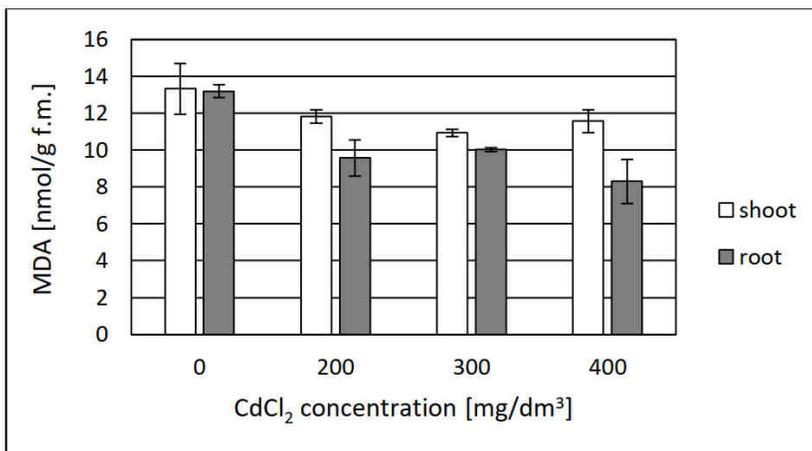


Fig. 4. The effect of different cadmium concentrations on malondialdehyde (MDA) content in *Beta vulgaris* plants (bars represent standard deviation); ANOVA, $p < 0.05$ (shoot), $p < 0.001$ (root)

Lipid peroxidation

Levels of lipid peroxidation in plant tissues of *B. vulgaris* were expressed as MDA content and were significantly lower in samples treated with Cd compared to the control (Fig. 4). However, no significant differences were found in any of the samples treated with Cd.

Photosynthetic pigments content

The effects of Cd on photosynthetic pigments content are shown in Figure 5. No significant differences were observed in pigments content between Cd-treated samples and the control.

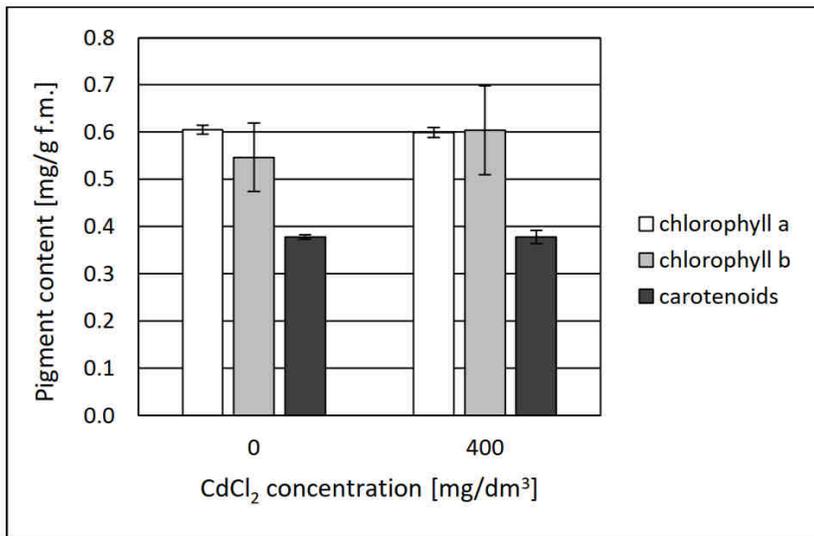


Fig. 5. The effect of different cadmium concentrations on photosynthetic pigments content in *Beta vulgaris* shoots (bars represent standard deviation); ANOVA, chl. a $p > 0.05$, chl. b $p > 0.05$, carotenoids $p > 0.05$

Cadmium content

The cadmium content in plant tissues depended on the concentration of CdCl₂ solution applied during the seed soaking process (Fig. 6). The seedlings not treated with cadmium had trace concentrations of Cd in both roots and shoots. The roots and shoots of control seedlings contained up to 0.39 mg Cd/kg d.m. (dry mass). Exposure of the seeds to excess Cd caused a rapid increase in Cd content in the plants. Maximum uptake (278.08 mg/kg d.m. in roots and 90.18 mg/kg d.m. in shoots) was noted in the 400 mg/dm³ treated plants. No significant differences were observed in root Cd content between the samples of 300 and 400 mg/dm³ CdCl₂ treatment. Furthermore, cadmium accumulated more in roots than in shoots (3.8 times more for the CdCl₂ concentrations 200 mg/dm³, 5 times more for 300 mg/dm³, 3 times more for 400 mg/dm³).

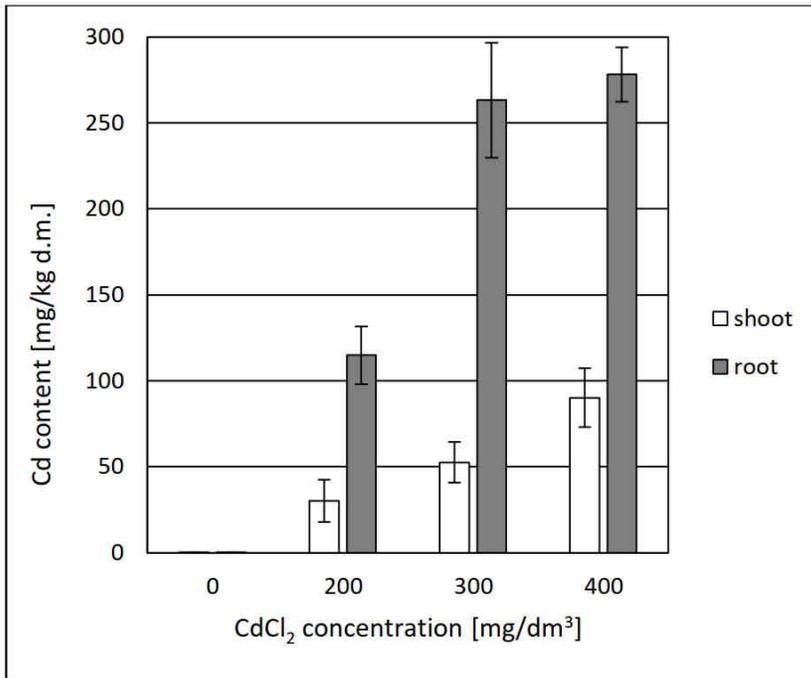


Fig. 6. The effect of different cadmium concentrations on cadmium content in *Beta vulgaris* plants (bars represent standard deviation); ANOVA, $p < 0.001$

Table 2
Correlation between cadmium content in *B. vulgaris* plants and MDA concentration, GPOX and SOD activity

Part of plant	Correlation coefficient <i>R</i>		
	Cd content/MDA	Cd content/GPOX activity	Cd content/SOD activity
Root	-0.79	0.68	-0.45
Shoot	-0.51	0.55	-0.32

The correlation between cadmium content in the roots and shoots of *B. vulgaris* and the concentration of MDA and antioxidant enzymes activity was investigated (Table 2). The analysis revealed a strong negative correlation between cadmium content and MDA concentration. In addition, a strong positive correlation between the plant's cadmium content and GPOX activity and a moderate negative correlation between its cadmium content and SOD activity were found.

Discussion

The main mechanism of cadmium phytotoxicity is the induction of oxidative stress. During Cd stress, systems capable of preventing excessive oxidation are induced or stimulated. These systems include some enzymes, such as superoxide dismutase (SOD) and peroxidases such as GPOX [24].

Superoxide dismutase constitutes the first line of defense against ROS by transforming O_2^- into O_2 and H_2O_2 [25]. In multiple studies, there have been varying outcomes on SOD

activity in plants exposed to Cd [26]. In this study, SOD activity in roots was stimulated by low Cd concentration (200 mg/dm³). However, high levels of Cd concentration diminished SOD activity compared to the control and 200 mg/dm³ concentration sample. These changes were similar to those in previous studies, where SOD activity in plants increased under low heavy metal concentrations and decreased under high concentrations, possibly as a result of a more severe degree of metal toxicity [27-31].

Peroxidases are important enzymes in plants that are able to lower oxidative stress induced by heavy metals. The main function of peroxidases is scavenging H₂O₂ in the cells. Guaiacol peroxidase utilizes H₂O₂ by oxidizing co-substrates such as phenolic compounds and/or antioxidants [32]. In this study, the GPOX activity in both roots and shoots significantly increased with higher cadmium concentration. This result was also obtained in other research, although GPOX activity varied based on Cd concentration and the plant species used [30, 33-36]. The higher activity of GPOX can explain the remarkably low concentration of MDA in *B. vulgaris* treated with Cd and the reduction in the lipid peroxidation level. These results indicated that GPOX in *B. vulgaris* played a more important role in ROS scavenging than SOD did. Furthermore, SOD and GPOX activity was approximately 50 % higher in roots compared to shoots. The roots also accumulated more cadmium than shoots (up to 5 times more for 300 mg/dm³ of CdCl₂). In most plant species, cadmium and other heavy metals mainly accumulate in roots [28, 35, 37-41]. The results suggest that increased accumulation of Cd in the roots of *B. vulgaris* stimulates GPOX activity, which protects cells from the negative effects of ROS.

In numerous other studies, chlorophyll and carotenoids have proven very sensitive to oxidative stress induced by Cd [35, 36, 42]. In the present study, however, there was no effect on photosynthetic pigments content in response to Cd, in the concentrations used. Similar results were observed in Cd-exposed *Vicia faba* [30].

In our study, there was no significant change in protein content under Cd stress (in the concentration range used). Similar findings were observed in garden cress, mustard and rapeseed, which could be due to the induction of synthesis of stress proteins [30, 42, 43].

Conclusion

Cadmium, in the concentration range used, did not show any significant effect on protein and photosynthetic pigments content in the roots and shoots of *Beta vulgaris*. Furthermore, the results of the present study revealed increased GPOX activity as cadmium concentration rose, while SOD activity was stimulated by low Cd concentration (200 mg/dm³) and diminished by high levels of Cd concentration. It can be concluded that GPOX in *B. vulgaris* played a more key role in ROS scavenging than SOD did and was able to reduce the level of lipid peroxidation in plants. Our results indicate reliable correlations between cadmium content and the values of oxidative stress biomarkers in plant tissues. The estimated correlation coefficients confirmed that the studied parameters depend on cadmium concentration.

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